



# STIC Search Report

**EIC 1700**

STIC Database Tracking Number: 192563

**TO:** Ben Sackey  
**Location:** REM 5B31  
**Art Unit :** 1626  
**June 16, 2006**

**Case Serial Number:** 10/734596

**From:** Kathleen Fuller  
**Location:** EIC 1700  
**REMSEN 4B28**  
**Phone:** 571/272-2505  
**Kathleen.Fuller@uspto.gov**

## Search Notes

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ACCESS DB # 192563

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Mrs. Franklin

Scientific and Technical Information Center

## SEARCH REQUEST FORM

Requester's Full Name: Ben Sackey Examiner #: 73489 Date: 6/10/86  
Art Unit: 1626 Phone Number: 2-0704 Serial Number: 101734,596  
Location (Bldg/Room#): REM 5B3 (Mailbox #): C 58 Results Format Preferred (circle): PAPER, DISK  
\*\*\*\*\*

To ensure an efficient and quality search, please attach a copy of the cover sheet, claims, and abstract or fill out the following:

Title of Invention: Process to make a conductive Copolymer of a fluorinated Polymer  
Inventors (please provide full names): Kerzhenko et al.

Earliest Priority Date: 12/13/82

Search Topic:

Please provide a detailed statement of the search topic, and describe as specifically as possible the subject matter to be searched. Include the elected species or structures, keywords, synonyms, acronyms, and registry numbers, and combine with the concept or utility of the invention. Define any terms that may have a special meaning. Give examples or relevant citations, authors, etc., if known.

\*For Sequence Searches Only\* Please include all pertinent information (parent, child, divisional, or issued patent numbers) along with the appropriate serial number.

A process to prepare a conductive fluorinated polymer composition wherein:

(a) an aqueous solution of an anilinium salt is mixed into an aqueous dispersion of a fluorinated polymer

(b) an oxidant for polymerizing the anilinium salt is added to the mixture of (a) to make a blend of doped polyaniiline fluorinated Polymer and resultant doped Polyaniiline fluorinated Polymer and resultant doped Polyaniiline

(c) by-products are removed to obtain a purified blend of fluorinated Polymer and doped Polyaniiline

(d) optionally, the purified fluorinated Polymer and doped Polyaniiline are mixed with an acid

(e) H<sub>2</sub>O is removed from purified fluorinated Polymer and doped Polyaniiline to obtain a powder.

40 min

SCIENTIFIC REFERENCE BR  
Sci & Tech Inf. Ctr

JUN 1 1986

Pat. & T.M. Office

Thanks

Sackey 10/734596 06/16/2006 Page 1

=> file reg  
FILE 'REGISTRY' ENTERED AT 15:38:57 ON 16 JUN 2006  
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STRUCTURE FILE UPDATES: 15 JUN 2006 HIGHEST RN 887970-41-4  
DICTIONARY FILE UPDATES: 15 JUN 2006 HIGHEST RN 887970-41-4

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\*\*\*\*\*  
\*  
\* The CA roles and document type information have been removed from \*  
\* the IDE default display format and the ED field has been added, \*  
\* effective March 20, 2005. A new display format, IDERL, is now \*  
\* available and contains the CA role and document type information. \*  
\*  
\*\*\*\*\*

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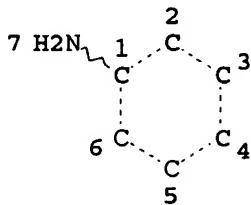
FILE COVERS 1907 - 16 Jun 2006 VOL 144 ISS 26  
FILE LAST UPDATED: 15 Jun 2006 (20060615/ED)

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This file contains CAS Registry Numbers for easy and accurate substance identification.

=> d que 132

L8 STR 1



NODE ATTRIBUTES:

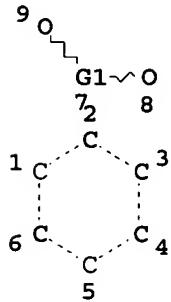
DEFAULT MLEVEL IS ATOM  
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I  
NUMBER OF NODES IS 7

STEREO ATTRIBUTES: NONE

L10 STR 2



24,660 polymers  
from 1 and 2

VAR G1=C/P/S/SE

NODE ATTRIBUTES:

DEFAULT MLEVEL IS ATOM  
DEFAULT ECLEVEL IS LIMITED

GRAPH ATTRIBUTES:

RSPEC I  
NUMBER OF NODES IS 9

STEREO ATTRIBUTES: NONE

L12 SCR 2043

L14 10633 SEA FILE=REGISTRY ABB=ON FLPO/PCT

L15 24660 SEA FILE=REGISTRY SSS FUL L8 AND L10 AND L12

L17 21544 SEA FILE=HCAPLUS ABB=ON L15

L18 8879 SEA FILE=HCAPLUS ABB=ON L17(L) PREP/RL

L19 82726 SEA FILE=HCAPLUS ABB=ON L14

L20 48 SEA FILE=HCAPLUS ABB=ON L18 AND L19

L21 10 SEA FILE=HCAPLUS ABB=ON L20 AND CONDUCT?

L22 166 SEA FILE=HCAPLUS ABB=ON L18 AND FLUOR? (2A) ?POLYMER?

L23 16 SEA FILE=HCAPLUS ABB=ON L22 AND CONDUCT?

L24 831 SEA FILE=HCAPLUS ABB=ON ?ANILIN? AND FLUOR? (2A) ?POLYMER?

L25 161 SEA FILE=HCAPLUS ABB=ON L24 AND CONDUCT?

L26 132 SEA FILE=HCAPLUS ABB=ON L25 AND POLYANILIN?

L27	32 SEA FILE=HCAPLUS ABB=ON	L26 AND (H2O OR WATER? OR AQUEOUS?)
L28	54 SEA FILE=HCAPLUS ABB=ON	L21 OR L23 OR L27
L29	43 SEA FILE=HCAPLUS ABB=ON	L28 AND (POLYMER? OR PLASTIC?)/SC, SX
L30	9 SEA FILE=HCAPLUS ABB=ON	L29 AND COMPOSITION?
L31	21 SEA FILE=HCAPLUS ABB=ON	L29 AND FILM?
L32	23 SEA FILE=HCAPLUS ABB=ON	L30 OR L31

=> d 132 bib abs ind hitstr 1-23

L32 ANSWER 1 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2006:439088 HCAPLUS  
 DN 144:469257  
 TI Conductive resin compositions and conductive  
 gel compositions with good film formability,  
 moldability, and processability  
 IN Tsukada, Yasuhiro; Furutani, Hiroyuki; Murakami, Mutsuaki; Yoshida,  
 Tatsushi  
 PA Kaneka Corporation, Japan  
 SO PCT Int. Appl., 51 pp.  
 CODEN: PIXXD2  
 DT Patent  
 LA Japanese  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2006049074	A1	20060511	WO 2005-JP19772	20051027
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, KE, KG, KM, KN, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, LY, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NG, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SM, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW				
	RW: AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IS, IT, LT, LU, LV, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG, BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM				
	JP 2006124615	A2	20060518	JP 2004-318437	20041101
	JP 2006152167	A2	20060615	JP 2004-346810	20041130
PRAI	JP 2004-318437	A	20041101		
	JP 2004-346810	A	20041130		
	JP 2005-73260	A	20050315		
AB	Title compns. contain a conductive polymer dispersed and/or dissolved in an ionic liquid. Thus, 4.02 g N-ethylimidazole was stirred in 20 mL DMF, 8.35 g Et p-toluenesulfonate was added therein and stirred for 23 h to give 1,3-diethylimidazolium p-toluenesulfonate, 10 mL of which was mixed with 0.50 g polypyrrole at 150° for 30 min, cooled to room temperature, filtered to remove undissolved polypyrrole, a filter paper soaked in the filtrate, and soaked in water to give a conductive article.				
CC	38-3 (Plastics Fabrication and Uses) Section cross-reference(s): 76				
ST	conductive resin compn gel film formability moldability processability; polypyrrole diethylimidazolium toluenesulfonate compn				
IT	Conducting polymers Electric conductors Gelation agents				

## Gels:

## Ionic liquids

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Molded plastics, uses

## Polyanilines

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Poly(arylenealkenylenes)

RL: TEM (Technical or engineered material use); USES (Uses)

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Fluoropolymers, uses

RL: MOA (Modifier or additive use); USES (Uses)

(geling agent; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Quinones

RL: TEM (Technical or engineered material use); USES (Uses)

(polymers; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Polyurethanes, uses

RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP (Preparation); USES (Uses)

(polyoxyalkylene-, geling agent; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Conducting polymers

(polypyrroles; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT Conducting polymers

(polythiophenes; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 30604-81-0P, Polypyrrole 51325-03-2P, Poly(1H-pyrrole-2,5-diyl)

126213-51-2P, 3,4-Ethylenedioxythiophene homopolymer 163359-60-2P

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 25233-30-1, Polyaniline

RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 4316-42-1, N-Butylimidazole

RL: RCT (Reactant); RACT (Reactant or reagent)

(conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 9042-77-7P, Polyethylene glycol-TDI copolymer

RL: IMF (Industrial manufacture); MOA (Modifier or additive use); PREP (Preparation); USES (Uses)

(geling agent; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 9011-17-0, Hexafluoropropylene-vinylidene fluoride copolymer

RL: MOA (Modifier or additive use); USES (Uses)

(geling agent; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 63458-90-2P 321842-70-0P 321842-72-2P 328090-25-1P 634922-90-0P

839672-88-7P 839672-91-2P 886220-74-2P  
 RL: IMF (Industrial manufacture); NUU (Other use, unclassified); PREP (Preparation); USES (Uses)

(ionic liquid; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 143314-16-3 145022-44-2 174501-64-5 174501-65-6 174899-82-2  
 304680-35-1 839672-85-4 868850-24-2 886220-75-3

RL: NUU (Other use, unclassified); USES (Uses)

(ionic liquid; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

IT 80-40-0, Ethyl p-toluenesulfonate 80-48-8, Methyl p-toluenesulfonate 616-47-7, N-Methylimidazole 1120-71-4, Propanesultone 3058-61-5 6192-52-5, p-Toluenesulfonic acid monohydrate 7098-07-9, N-Ethylimidazole 26158-00-9

RL: RCT (Reactant); RACT (Reactant or reagent)

(reactant in ionic liquid preparation; conductive resin compns. containing ionic liquid with good film formability, moldability, and processability)

RE.CNT 16 THERE ARE 16 CITED REFERENCES AVAILABLE FOR THIS RECORD  
 ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 2 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2005:671915 HCAPLUS

DN 143:176212

TI Electrolyte composition, solid electrolyte membrane, and its use in polymer electrolyte fuel cell

IN Nawarage, Florence Cooley

PA Fujitsu Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 52 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2005200441	A2	20050728	JP 2004-5003	20040113
PRAI JP 2004-5003		20040113		

AB The composition contains sulfo-containing polymers having defined ether units, benzoxazole units, imide units, and/or benzothiazole units. The membrane is obtained by energy ray irradiation to and/or heat treatment of the composition. The fuel cell uses the membrane. The membrane is scarcely deteriorated in strong acid atmospheric and shows low MeOH crossover and high proton conductivity

IC ICM C08G065-34

ICS C08F299-02; C08G059-00; C08G073-10; C08G073-22; C08G075-32; H01B001-06; H01M008-02; H01M008-10

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
 Section cross-reference(s): 38

ST sulfo polymer ether benzoxazole imide benzothiazole unit; solid electrolyte membrane polymer electrolyte fuel cell

IT Polyethers, uses

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (polybenzobisthiazole-, sulfo-containing; sulfo-containing polymers with ether,

benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Polyethers, uses

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(polybenzoxazole-, sulfo-containing; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Polysulfones, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyether-, sulfo-containing; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Polybenzoxazoles

Polyimides, uses

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyether-, sulfo-containing; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Polyethers, uses

RL: IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyimide-, sulfo-containing; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Fuel cells

(polymer electrolyte; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Polyethers, uses

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polysulfone-, sulfo-containing; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Ionic conductors

(protonic; sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT Fuel cell electrolytes

(sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT 861001-39-0P 861001-40-3P 861001-41-4P 861001-42-5P 861001-43-6P  
861001-44-7P

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT 1858-67-9P 13036-02-7P, Dimethyl 5-hydroxyisophthalate 36637-44-2P,  
4-(2-Tetrahydropyranloxy)phenyl magnesium bromide 145784-93-6P  
861001-17-4P 861001-18-5P 861001-19-6P 861001-20-9P 861001-21-0P  
861001-22-1P 861001-23-2P 861001-24-3P 861001-25-4P 861001-26-5P  
861001-27-6P 861001-28-7P 861001-29-8P 861001-30-1P 861001-31-2P  
861001-32-3P 861001-33-4P 861001-34-5P 861001-35-6P 861001-36-7P  
861001-37-8P

RL: IMF (Industrial manufacture); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent) (sulfo-containing polymers with ether, benzoxazole, imide, and/or benzothiazole units for solid electrolyte membrane in polymer electrolyte fuel cell)

IT 861001-45-8P 861001-46-9P 861001-47-0P 861001-48-1P 861001-49-2P

861001-50-5P 861001-51-6P 861001-52-7P 861001-53-8P 861001-54-9P  
 861001-55-0P 861001-56-1P 861001-57-2P 861001-58-3P  
 861001-59-4P 861001-60-7P 861001-61-8P  
 861001-62-9DP, compds. with epichlorohydrin, polymers with  
 bisphenol A diglycidyl ether 861001-63-0DP, compds. with  
 epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-63-0P 861001-65-2P 861001-66-3P  
 861001-67-4P 861001-68-5P 861001-69-6P 861001-70-9DP  
 , compds. with epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-71-0P 861001-72-1P 861001-73-2P 861001-74-3P  
 861001-76-5P 861001-78-7P 861001-79-8DP, compds. with  
 epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-81-2DP, compds. with epichlorohydrin, polymers with  
 bisphenol A diglycidyl ether  
 RL: IMF (Industrial manufacture); TEM (Technical or engineered material  
 use); PREP (Preparation); USES (Uses)  
 (sulfo-containing polymers with ether, benzoxazole, imide, and/or  
 benzothiazole units for solid electrolyte membrane in polymer  
 electrolyte fuel cell)

IT 106-44-5, p-Cresol, reactions 108-77-0, Cyanuric chloride 124-73-2,  
 1,2-Dibromotetrafluoroethane 349-88-2, 4-Fluorobenzenesulfonyl chloride  
 350-46-9, 4-Fluoronitrobenzene 352-34-1, 4-  
 Fluoroiodobenzene 375-50-8, 1,4-Diiodoperfluorobutane  
 500-99-2, 3,5-Dimethoxyphenol 618-83-7, 5-Hydroxyisophthalic acid  
 917-54-4, Methylolithium 999-97-3, Hexamethyldisilazane 1194-02-1, 4-  
 Fluorobenzonitrile 2039-82-9, 4-Bromostyrene 7446-09-5, Sulfur  
 dioxide, reactions 108534-47-0, 4-tert-Butyldimethylsilyloxyphenol  
 861001-38-9

RL: RCT (Reactant); RACT (Reactant or reagent)  
 (sulfo-containing polymers with ether, benzoxazole, imide, and/or  
 benzothiazole units for solid electrolyte membrane in polymer  
 electrolyte fuel cell)

IT 861001-57-2P 861001-58-3P 861001-60-7P  
 861001-61-8P 861001-62-9DP, compds. with  
 epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-63-0DP, compds. with epichlorohydrin, polymers with  
 bisphenol A diglycidyl ether 861001-63-0P 861001-65-2P  
 861001-66-3P 861001-68-5P 861001-70-9DP,  
 compds. with epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-71-0P 861001-73-2P 861001-79-8DP,  
 compds. with epichlorohydrin, polymers with bisphenol A diglycidyl ether  
 861001-81-2DP, compds. with epichlorohydrin, polymers with  
 bisphenol A diglycidyl ether  
 RL: IMF (Industrial manufacture); TEM (Technical or engineered material  
 use); PREP (Preparation); USES (Uses)  
 (sulfo-containing polymers with ether, benzoxazole, imide, and/or  
 benzothiazole units for solid electrolyte membrane in polymer  
 electrolyte fuel cell)

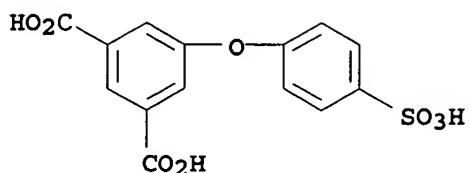
RN 861001-57-2 HCPLUS

CN 1,3-Benzenedicarboxylic acid, 5-(4-sulfophenoxy)-, tripotassium salt,  
 polymer with 1,4-benzenedicarboxylic acid and 3,3'-diamino[1,1'-biphenyl]-  
 4,4'-diol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-21-0

CMF C14 H10 O8 S . 3 K

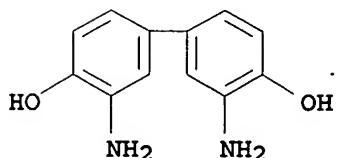


● 3 K

CM 2

CRN 4194-40-5

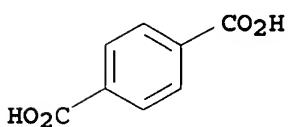
CMF C12 H12 N2 O2



CM 3

CRN 100-21-0

CMF C8 H6 O4



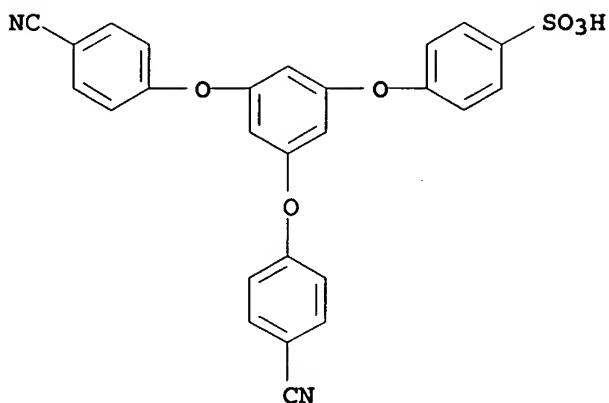
RN 861001-58-3 HCAPLUS

CN Benzenesulfonic acid, 4-[3,5-bis(4-cyanophenoxy)phenoxy]-, monopotassium salt, polymer with 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX NAME)

CM 1

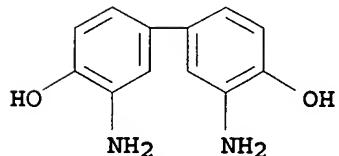
CRN 861001-22-1

CMF C26 H16 N2 O6 S . K



● K

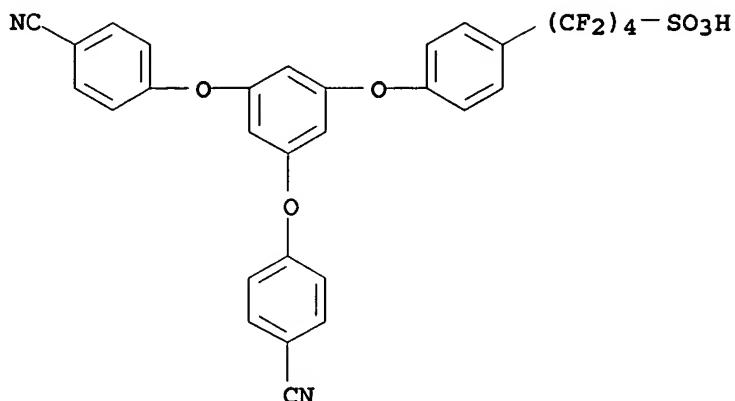
CM 2

CRN 4194-40-5  
CMF C12 H12 N2 O2

RN 861001-60-7 HCAPLUS  
CN 1,4-Benzenedicarboxylic acid, polymer with 4-[3,5-bis(4-cyanophenoxy)phenoxy]- $\alpha,\alpha,\beta,\beta,\gamma,\gamma,\delta$ ,  
.8-octafluorobenzenebutanesulfonic acid monopotassium salt and  
4,4'-diamino[1,1'-biphenyl]-3,3'-diol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-30-1  
CMF C30 H16 F8 N2 O6 S . K

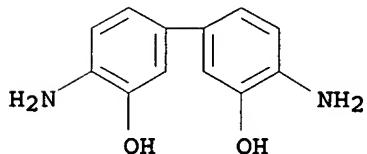


● K

CM 2

CRN 2373-98-0

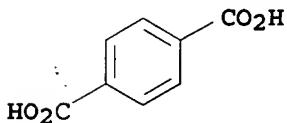
CMF C12 H12 N2 O2



CM 3

CRN 100-21-0

CMF C8 H6 O4



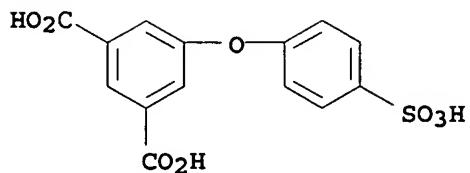
RN 861001-61-8 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-hydroxy-, polymer with  
 3,3'-diamino[1,1'-biphenyl]-4,4'-diol, 2,2'-(1-methylethylidene)bis(4,1-  
 phenyleneoxymethylene)bis[oxirane] and 5-(4-sulfophenoxy)-1,3-  
 benzenedicarboxylic acid tripotassium salt (9CI) (CA INDEX NAME)

CM 1

CRN 861001-21-0

CMF C14 H10 O8 S . 3 K

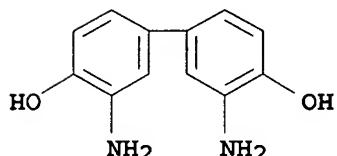


● 3 K

CM 2

CRN 4194-40-5

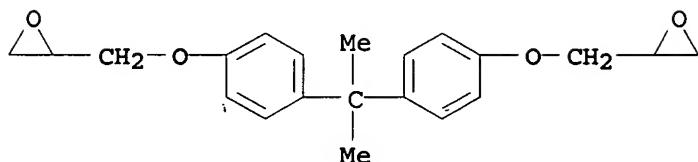
CMF C12 H12 N2 O2



CM 3

CRN 1675-54-3

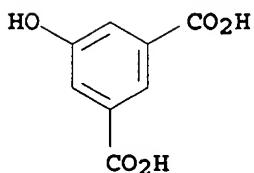
CMF C21 H24 O4



CM 4

CRN 618-83-7

CMF C8 H6 O5



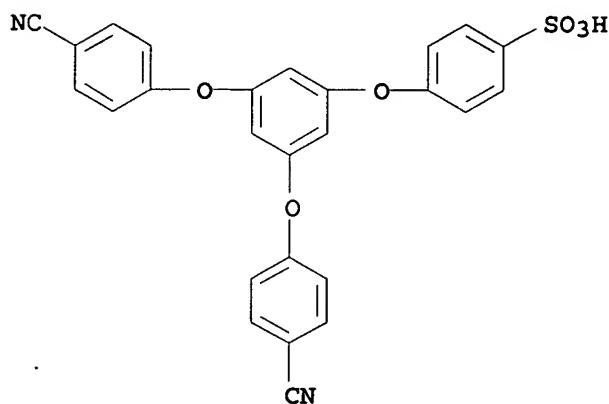
RN 861001-62-9 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-hydroxy-, polymer with  
 4-[3,5-bis(4-cyanophenoxy)phenoxy]benzenesulfonic acid monopotassium salt  
 and 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-22-1

CMF C26 H16 N2 O6 S . K

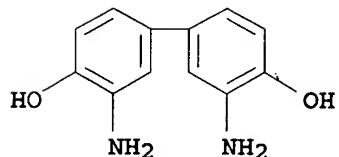


● K

CM 2

CRN 4194-40-5

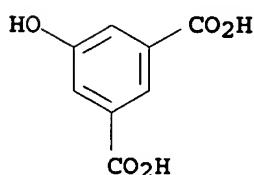
CMF C12 H12 N2 O2



CM 3

CRN 618-83-7

CMF C8 H6 O5



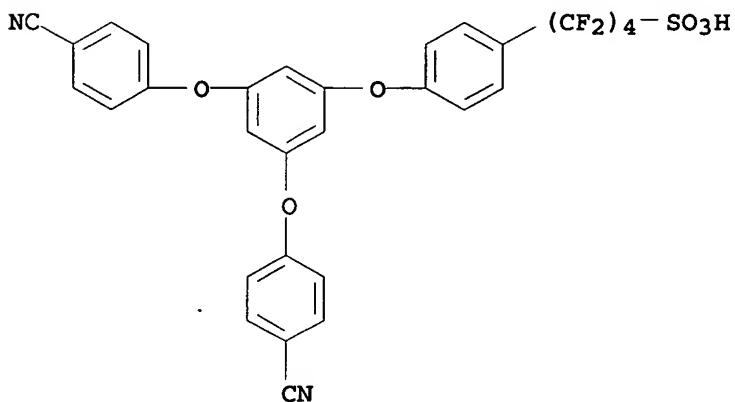
RN 861001-63-0 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-hydroxy-, polymer with  
 4-[3,5-bis(4-cyanophenoxy)phenoxy]- $\alpha,\alpha,\beta,\beta,\gamma,\gamma$ ,  
 gamma., $\delta,\delta$ -octafluorobenzenebutanesulfonic acid monopotassium  
 salt and 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-30-1

CMF C30 H16 F8 N2 O6 S . K

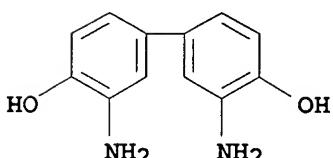


● K

CM 2

CRN 4194-40-5

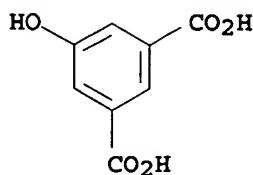
CMF C12 H12 N2 O2



CM 3

CRN 618-83-7

CMF C8 H6 O5



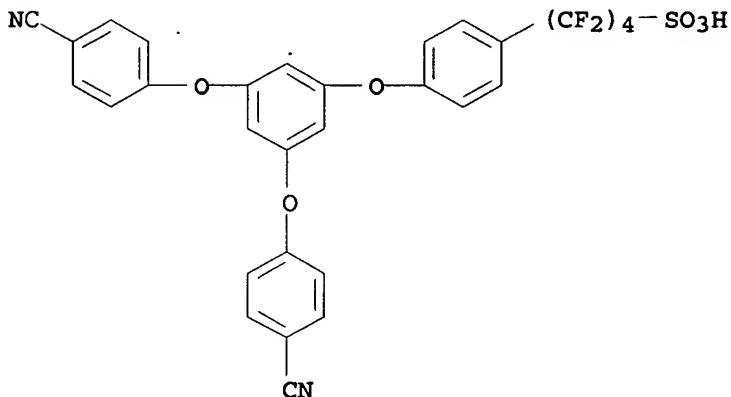
RN 861001-63-0 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-hydroxy-, polymer with  
 4-[3,5-bis(4-cyanophenoxy)phenoxy]- $\alpha,\alpha,\beta,\beta,\gamma,\gamma$ ,  
 gamma,,8,8-octafluorobenzenebutanesulfonic acid monopotassium  
 salt and 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-30-1

CMF C30 H16 F8 N2 O6 S . K

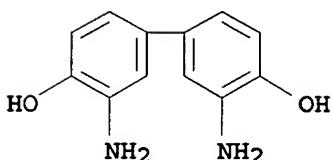


● K

CM 2

CRN 4194-40-5

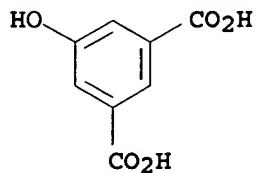
CMF C12 H12 N2 O2



CM 3

CRN 618-83-7

CMF C8 H6 O5



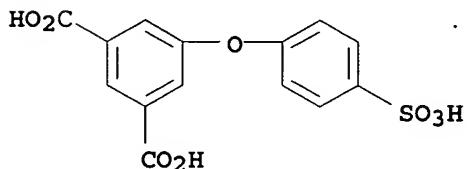
RN 861001-65-2 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-(4-sulfophenoxy)-, tripotassium salt, polymer with 1,4-benzenedicarboxylic acid and 3,3'-diamino[1,1'-biphenyl]-4,4'-dithiol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-21-0

CMF C14 H10 O8 S . 3 K

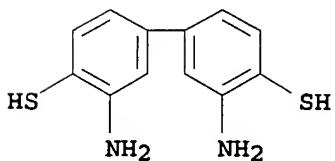


● 3 K

CM 2

CRN 51818-67-8

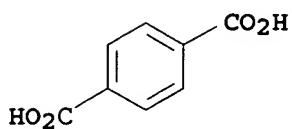
CMF C12 H12 N2 S2



CM 3

CRN 100-21-0

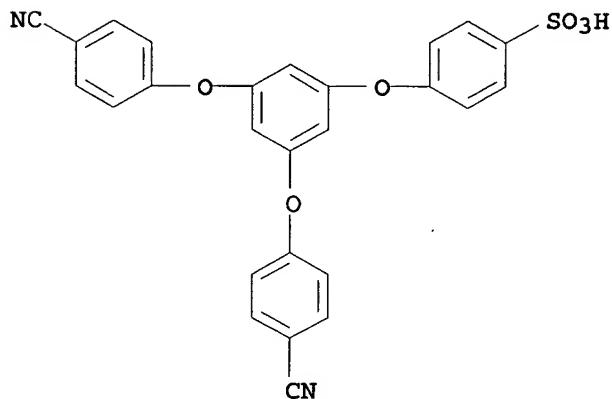
CMF C8 H6 O4



RN 861001-66-3 HCAPLUS  
 CN Benzenesulfonic acid, 4-[3,5-bis(4-cyanophenoxy)phenoxy]-, monopotassium salt, polymer with 3,3'-diamino[1,1'-biphenyl]-4,4'-dithiol (9CI) (CA INDEX NAME)

CM 1

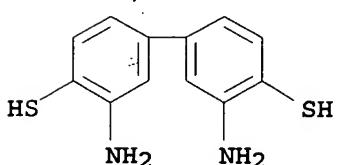
CRN 861001-22-1  
 CMF C26 H16 N2 O6 S . K



● K

CM 2

CRN 51818-67-8  
 CMF C12 H12 N2 S2

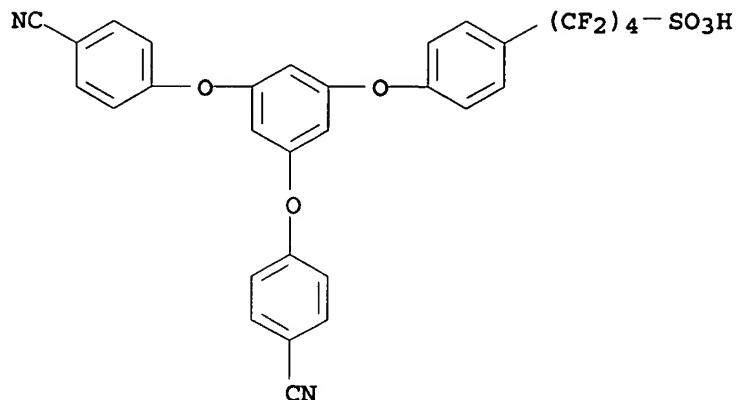


RN 861001-68-5 HCAPLUS  
 CN Benzenebutanesulfonic acid, 4-[3,5-bis(4-cyanophenoxy)phenoxy]- $\alpha,\alpha,\beta,\beta,\gamma,\gamma,\delta,\delta$ -octafluoro-, monopotassium salt, polymer with 3,3'-diamino[1,1'-biphenyl]-4,4'-dithiol (9CI) (CA INDEX NAME)

CM 1

CRN 861001-30-1

CMF C30 H16 F8 N2 O6 S . K

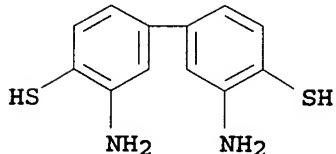


● K

CM 2

CRN 51818-67-8

CMF C12 H12 N2 S2



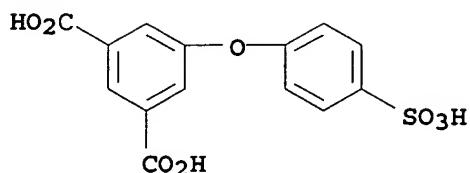
RN 861001-70-9 HCAPLUS

CN 1,3-Benzenedicarboxylic acid, 5-hydroxy-, polymer with  
3,3'-diamino[1,1'-biphenyl]-4,4'-diol and 5-(4-sulfophenoxy)-1,3-  
benzenedicarboxylic acid tripotassium salt (9CI) (CA INDEX NAME)

CM 1

CRN 861001-21-0

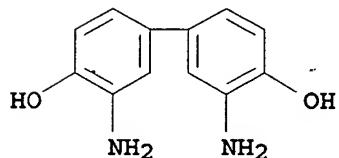
CMF C14 H10 O8 S . 3 K



● 3 K

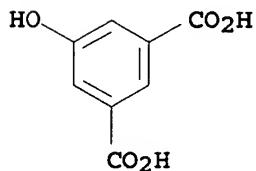
CM 2

CRN 4194-40-5  
CMF C12 H12 N2 O2



CM 3

CRN 618-83-7  
CMF C8 H6 O5

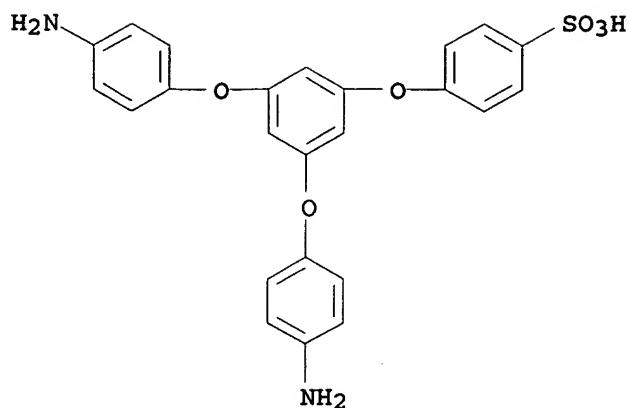


RN 861001-71-0 HCAPLUS

CN Benzenesulfonic acid, 4-[3,5-bis(4-aminophenoxy)phenoxy]-, monopotassium salt, polymer with [2]benzopyrano[6,5,4-def] [2]benzopyran-1,3,6,8-tetrone (9CI) (CA INDEX NAME)

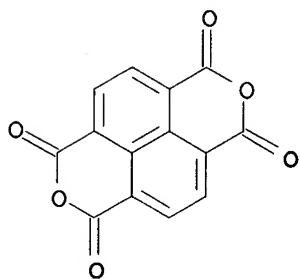
CM 1

CRN 861001-24-3  
CMF C24 H20 N2 O6 S . K



● K

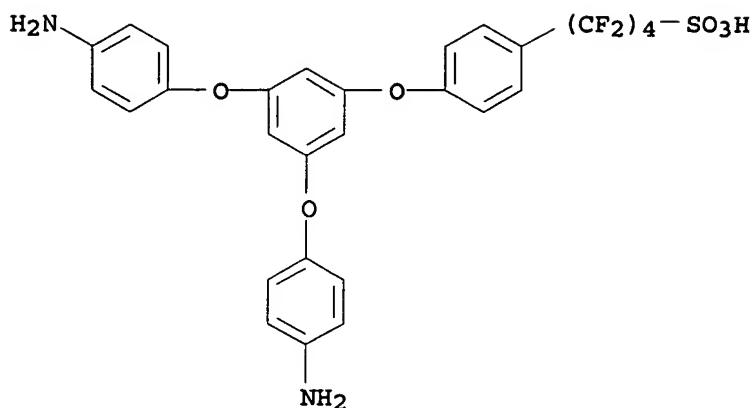
CM 2

CRN 81-30-1  
CMF C14 H4 O6

RN 861001-73-2 HCAPLUS  
CN Benzenebutanesulfonic acid, 4-[3,5-bis(4-aminophenoxy)phenoxy]-  
α,α,β,β,γ,γ,δ,δ-octafluoro-,  
monoammonium salt, polymer with [2]benzopyrano[6,5,4-def][2]benzopyran-  
1,3,6,8-tetrone (9CI) (CA INDEX NAME)

CM 1

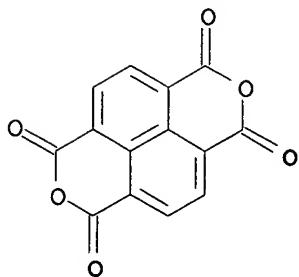
CRN 861001-32-3  
CMF C28 H20 F8 N2 O6 S . H3 N



● NH<sub>3</sub>

CM 2

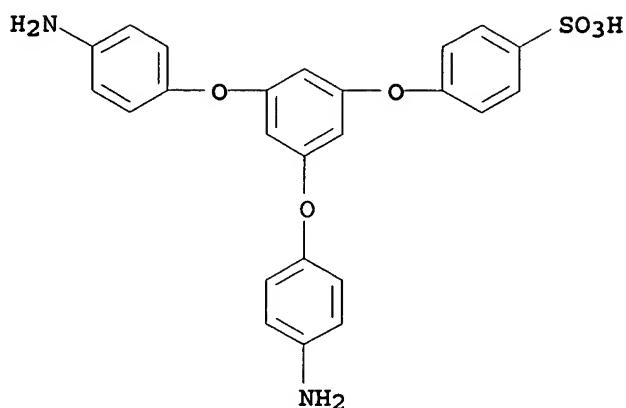
CRN 81-30-1  
CMF C14 H4 O6



RN 861001-79-8 HCAPLUS  
CN Benzenesulfonic acid, 4-[3,5-bis(4-aminophenoxy)phenoxy]-, monopotassium salt, polymer with [2]benzopyrano[6,5,4-def] [2]benzopyran-1,3,6,8-tetrone and 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX NAME)

CM 1

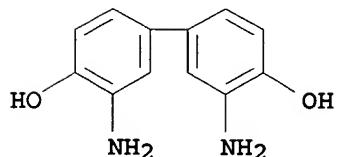
CRN 861001-24-3  
CMF C24 H20 N2 O6 S . K



● K

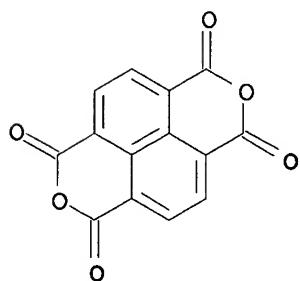
CM 2

CRN 4194-40-5  
 CMF C12 H12 N2 O2



CM 3

CRN 81-30-1  
 CMF C14 H4 O6



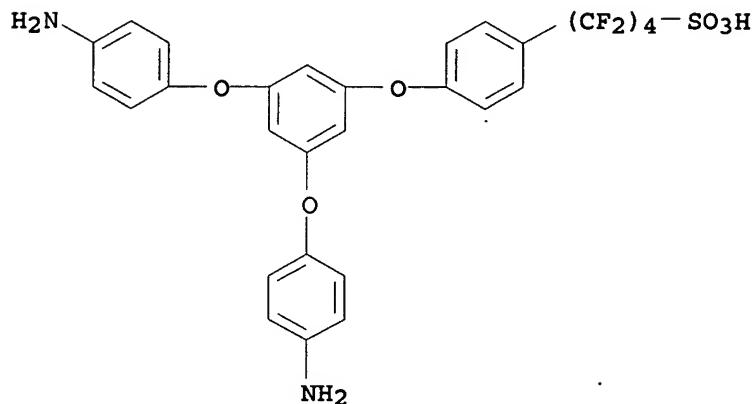
RN 861001-81-2 HCAPLUS  
 CN Benzenebutanesulfonic acid, 4-[3,5-bis(4-aminophenoxy)phenoxy]-  
 $\alpha,\alpha,\beta,\beta,\gamma,\gamma,\delta,\delta$ -octafluoro-,  
 monopotassium salt, polymer with [2]benzopyrano[6,5,4-def] [2]benzopyran-

1,3,6,8-tetrone and 3,3'-diamino[1,1'-biphenyl]-4,4'-diol (9CI) (CA INDEX  
NAME)

CM 1

CRN 861001-80-1

CMF C28 H20 F8 N2 O6 S . K

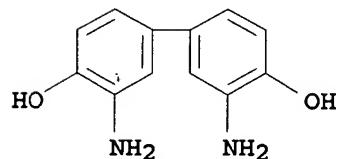


● K

CM 2

CRN 4194-40-5

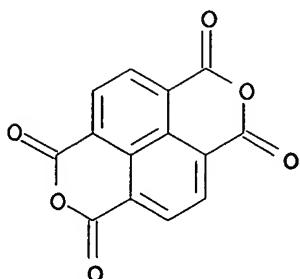
CMF C12 H12 N2 O2



CM 3

CRN 81-30-1

CMF C14 H4 O6



L32 ANSWER 3 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2005:179359 HCAPLUS  
 DN 142:269523  
 TI Plastic substrates and their electroconductive laminates for display devices, solar cells, and touch panels  
 IN Kuma, Takuya; Moriyama, Hideki; Tsukuda, Akimitsu; Kitajima, Hodaka  
 PA Toray Industries, Inc., Japan  
 SO Jpn. Kokai Tokkyo Koho, 23 pp.  
 CODEN: JKXXAF  
 DT Patent  
 LA Japanese  
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2005054173	A2	20050303	JP 2004-208193	20040715
PRAI JP 2003-198963	A	20030718		

AB The plastic substrates show  $T_g \geq 260^\circ$  and  $<600^\circ$ , light transmittance at 400 nm 60-100%, total light transmittance at 450-700 nm 80-100%, and pencil hardness  $\geq H$ . The films show good mech. properties and bending crack resistance.  
 IC ICM C08J005-18  
 ICS B32B007-02; B32B027-34; H05K001-03; C23C014-08; C08L077-10  
 CC 74-13 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)  
 Section cross-reference(s): 38, 52, 76  
 ST arom polyamide substrate electroconductive laminate; liq crystal display solar cell touch panel; aminophenyl fluorene terephthaloyl chloride polyamide substrate  
 IT Liquid crystal displays  
 Optical imaging devices  
 Solar cells  
 (aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)  
 IT Laminated plastics, uses  
 RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
 (aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)  
 IT Polyamides, preparation  
 RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (cardo; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)  
 IT Transparent films  
 (elec. conductive; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch

panels)

IT Electric conductors

(films, transparent; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Polyamides, preparation

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (fluorine-containing, cardo; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Polyamides, preparation

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (fluorine-containing; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Fluoropolymers, preparation

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamide-, cardo; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Fluoropolymers, preparation

Polysulfones, preparation  
RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamide-; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Cardo polymers

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamides, fluorine-containing; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Cardo polymers

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polyamides; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Polyamides, preparation

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses) (polysulfone-; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT Electric switches

(touch panels; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT 26285-32-5P 29931-02-0P, 9,9-Bis(4-aminophenyl)fluorene-terephthaloyl dichloride copolymer 37372-31-9P 65681-31-4P, 9,9-(4-Aminophenyl)fluorene-isophthaloyl chloride-terephthaloyl chloride copolymer 65722-41-0P 96194-37-5P 96194-42-2P 99634-81-8P, 3,3'-Diaminodiphenyl sulfone-4,4'-diaminodiphenyl sulfone-terephthaloyl chloride copolymer 686777-58-2P 686777-69-5P 688046-12-0P 845816-35-5P 845816-36-6P 846603-87-0P 846603-91-6P

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT 50926-11-9, ITO

RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(electroconductive layer; aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

IT 99634-81-8P, 3,3'-Diaminodiphenyl sulfone-4,4'-diaminodiphenyl sulfone-terephthaloyl chloride copolymer

RL: DEV (Device component use); IMF (Industrial manufacture); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(aromatic polyamide substrates and their electroconductive laminates for display devices, solar cells, and touch panels)

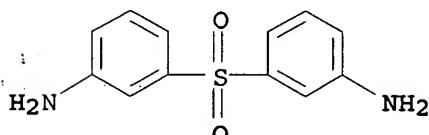
RN 99634-81-8 HCAPLUS

CN 1,4-Benzenedicarbonyl dichloride, polymer with 3,3'-sulfonylbis[benzenamine] and 4,4'-sulfonylbis[benzenamine] (9CI) (CA INDEX NAME)

CM 1

CRN 599-61-1

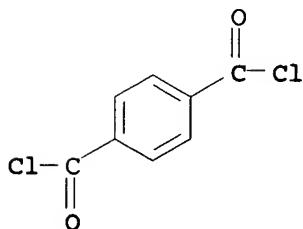
CMF C12 H12 N2 O2 S



CM 2

CRN 100-20-9

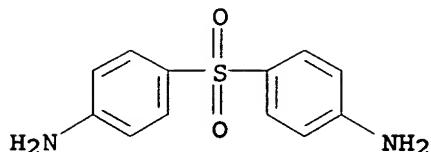
CMF C8 H4 Cl2 O2



CM 3

CRN 80-08-0

CMF C12 H12 N2 O2 S



L32 ANSWER 4 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2005:14475 HCAPLUS  
 DN 142:117631  
 TI Polymer composition for encapsulation of electrode particles  
 IN Gozdz, Antoni S.; Loxley, Andrew L.; Pullen, Anthony E.  
 PA A123 Systems, Inc., USA  
 SO PCT Int. Appl., 47 pp.  
 CODEN: PIXXD2

DT Patent  
 LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	WO 2005000956	A2	20050106	WO 2004-US20393	20040623
	WO 2005000956	A3	20051110		
	W: AE, AG, AL, AM, AT, AU, AZ, BA, BB, BG, BR, BW, BY, BZ, CA, CH, CN, CO, CR, CU, CZ, DE, DK, DM, DZ, EC, EE, EG, ES, FI, GB, GD, GE, GH, GM, HR, HU, ID, IL, IN, IS, JP, KE, KG, KP, KR, KZ, LC, LK, LR, LS, LT, LU, LV, MA, MD, MG, MK, MN, MW, MX, MZ, NA, NI, NO, NZ, OM, PG, PH, PL, PT, RO, RU, SC, SD, SE, SG, SK, SL, SY, TJ, TM, TN, TR, TT, TZ, UA, UG, US, UZ, VC, VN, YU, ZA, ZM, ZW RW: BW, GH, GM, KE, LS, MW, MZ, NA, SD, SL, SZ, TZ, UG, ZM, ZW, AM, AZ, BY, KG, KZ, MD, RU, TJ, TM, AT, BE, BG, CH, CY, CZ, DE, DK, EE, ES, FI, FR, GB, GR, HU, IE, IT, LU, MC, NL, PL, PT, RO, SE, SI, SK, TR, BF, BJ, CF, CG, CI, CM, GA, GN, GQ, GW, ML, MR, NE, SN, TD, TG				
PRAI	US 2005034993	A1	20050217	US 2004-876179	20040623
AB	US 2003-480535P	P	20030623		
IC	Comps. and methods are provided for coating electroactive particles. Coating materials include a conductive component and a low refractive index component. Coatings are provided in which the conductive and low refractive index components are linked and/or do not form phases having length scales .gtorsim.0.25 $\mu$ m. Coatings are provided in which the components are contained in sequential layers.				
CC	52-2 (Electrochemical, Radiational, and Thermal Energy Technology) Section cross-reference(s): 35, 42, 76				
ST	encapsulation electrode particle low refractive index conductive polymer; electropolymer radical polymer poly thiophene acrylic fluoroalkyl conductive polymer; electrochem cell encapsulated electrode oxide polymd polythiophene acrylic graft				
IT	Polymers, uses RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses) (block, encapsulating polymer; polymer composition for encapsulation of electrode particles)				
IT	Electric current (collector, coated; polymer composition for encapsulation of electrode particles)				
IT	Bond				

(covalent, between coating components; polymer composition for encapsulation of electrode particles)

IT Phase separation  
(domain size  $\leq 0.25 \mu\text{m}$ ; polymer composition for encapsulation of electrode particles)

IT Electric apparatus

Polymerization  
(electrochem.; polymer composition for encapsulation of electrode particles)

IT Polyacetylenes, uses

Polyanilines

Polyphenyls

RL: TEM (Technical or engineered material use); USES (Uses)  
(encapsulating polymer; polymer composition for encapsulation of electrode particles)

IT Electrodes  
(encapsulation of particulate materials for; polymer composition for encapsulation of electrode particles)

IT Fluoropolymers, uses  
RL: PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); PYP (Physical process); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); PROC (Process); USES (Uses)  
(fluoroalkyl and fluoroaryl groups; polymer composition for encapsulation of electrode particles)

IT Ethers, preparation  
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
(fluoroalkyl, ethers with hydroxythiophenes; polymer composition for encapsulation of electrode particles)

IT Polymers, uses  
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)  
(graft, encapsulating polymer; polymer composition for encapsulation of electrode particles)

IT Bond  
(ionic, between coating components; polymer composition for encapsulation of electrode particles)

IT Films  
(multilayer; polymer composition for encapsulation of electrode particles)

IT Refractive index  
(of coating polymer; polymer composition for encapsulation of electrode particles)

IT Electric conductivity  
(of polymers and encapsulated oxides; polymer composition for encapsulation of electrode particles)

IT Polymerization  
(oxidative coupling; polymer composition for encapsulation of electrode particles)

IT Anodes

Cathodes

Coating materials

Conducting polymers

Electrolytes

Encapsulation

Etherification

Fluorination

Oxidizing agents

Solvents

- (polymer composition for encapsulation of electrode particles)
- IT Acrylic polymers, uses  
RL: TEM (Technical or engineered material use); USES (Uses)  
(polymer composition for encapsulation of electrode particles)
- IT Conducting polymers  
(polypyrroles, encapsulating polymer, alkyl, ether, thioether, ester, thioester, amine, amide, and benzyl derivs.; polymer composition for encapsulation of electrode particles)
- IT Conducting polymers  
(polythiophenes, encapsulating polymers, alkyl, ether, thioether, alkylenedioxy-, ester, thioester, amine, amide, and benzyl derivs.; polymer composition for encapsulation of electrode particles)
- IT Force  
(repulsive, of polymers to MCMBs; polymer composition for encapsulation of electrode particles)
- IT Coating process  
(spray; polymer composition for encapsulation of electrode particles)
- IT Glass substrates  
(substrate for electrode for electropolymer.; polymer composition for encapsulation of electrode particles)
- IT Polymerization  
(vapor-deposition; polymer composition for encapsulation of electrode particles)
- IT 627528-57-8P  
RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
(EDOT-F monomer; polymer composition for encapsulation of electrode particles)
- IT 7440-44-0, Carbon, uses  
RL: PRP (Properties); TEM (Technical or engineered material use); USES (Uses)  
(MCMB; polymer composition for encapsulation of electrode particles)
- IT 820958-17-6P  
RL: SPN (Synthetic preparation); PREP (Preparation)  
(PrODOT-F monomer; polymer composition for encapsulation of electrode particles)
- IT 52627-24-4, Lithium cobalt oxide 162684-16-4, Lithium manganese nickel oxide  
RL: PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process)  
(electrode material, encapsulation of; polymer composition for encapsulation of electrode particles)
- IT 820958-24-5P  
RL: PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)  
(oxide encapsulant film; polymer composition for encapsulation of electrode particles)
- IT 7646-69-7, Sodium hydride 7681-65-4, Copper iodide (CuI) 7705-08-0, Ferric chloride, uses 7727-54-0, Ammonium persulfate 10421-48-4, Ferric nitrate 13537-24-1, Ferric perchlorate  
RL: CAT (Catalyst use); USES (Uses)  
(polymer composition for encapsulation of electrode particles)
- IT 312619-41-3  
RL: CAT (Catalyst use); MOA (Modifier or additive use); PEP (Physical, engineering or chemical process); PYP (Physical process); PROC (Process); USES (Uses)

(polymer composition for encapsulation of electrode particles)

IT 7440-47-3, Chromium, uses 7440-57-5, Gold, uses  
 RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)

(polymer composition for encapsulation of electrode particles)

IT 64-17-5, Ethanol, uses 67-56-1, Methanol, uses 67-64-1, Acetone, uses 67-66-3, Chloroform, uses 75-05-8, Acetonitrile, uses 75-09-2, Methylene chloride, uses 141-78-6, Ethyl acetate, uses 7732-18-5, Water, uses  
 RL: NUU (Other use, unclassified); USES (Uses)

(polymer composition for encapsulation of electrode particles)

IT 28825-23-2, Poly(hexafluoroisopropylmethacrylate) 104934-51-2, Poly(3-octylthiophene)  
 RL: PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PYP (Physical process); PROC (Process); USES (Uses)

(polymer composition for encapsulation of electrode particles)

IT 142214-55-9P 153634-17-4P 627528-58-9P 820958-20-1P  
 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PUR (Purification or recovery); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(polymer composition for encapsulation of electrode particles)

IT 820958-29-0P  
 RL: PEP (Physical, engineering or chemical process); PUR (Purification or recovery); PYP (Physical process); SPN (Synthetic preparation); PREP (Preparation); PROC (Process)

(polymer composition for encapsulation of electrode particles)

IT 155090-83-8, Baytron P  
 RL: PRP (Properties); RCT (Reactant); RACT (Reactant or reagent)

(polymer composition for encapsulation of electrode particles)

IT 155090-83-8DP, Baytron P, fluorinated  
 RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)

(polymer composition for encapsulation of electrode particles)

IT 307-30-2 872-31-1, 3-Bromothiophene 920-46-7, Methacryloyl chloride 7782-41-4, Fluorine, reactions 13781-67-4, 3-Thiopheneethanol 820958-26-7  
 RL: RCT (Reactant); RACT (Reactant or reagent)

(polymer composition for encapsulation of electrode particles)

IT 142214-54-8P  
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)

(polymer composition for encapsulation of electrode particles)

IT 153634-15-2P  
 RL: SPN (Synthetic preparation); PREP (Preparation)

(polymer composition for encapsulation of electrode particles)

IT 9002-84-0, Polytetrafluoroethylene 9003-53-6, Polystyrene  
 RL: TEM (Technical or engineered material use); USES (Uses)

(polymer composition for encapsulation of electrode particles)

L32 ANSWER 5 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2004:533195 HCAPLUS  
 DN 141:79294  
 TI Semiconductor compositions and electrophotographic apparatus parts using them with excellent heat, moisture, and voltage resistance  
 IN Yoshikawa, Hitoshi; Iinuma, Sumio  
 PA Tokai Rubber Industries, Ltd., Japan  
 SO Jpn. Kokai Tokkyo Koho, 35 pp.  
 CODEN: JKXXAF  
 DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2004184513	A2	20040702	JP 2002-348351	20021129
PRAI	JP 2002-348351		20021129		
OS	MARPAT 141:79294				
AB	The compns., development rolls for electrophotog., contain elec. conductive polymers (A) having surfactant structures (sulfonic acid group-containing naphthalene or anthracene structures, preferably) and showing solubility to PhMe or Me Et ketone $\geq 20\%$ and solubility to water $< 3\%$ and binder polymers (B), thus improving compatibility of them.				
IC	ICM G03G015-08 ICS C08L101-00; F16C013-00; G03G015-02; G03G015-16; H01B001-20				
CC	74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes) Section cross-reference(s): 38, 76				
ST	elec conductor polymer soly moisture resistance; sulfonic surfactant polyaniline polyester blend compatibility; electrophotog development roll semiconductor heat resistance				
IT	Acrylic rubber RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Denka ER 7300T, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)				
IT	Urethane rubber, uses RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (Elastollan 1040, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)				
IT	Butadiene rubber, uses RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (JSR-BR 1220NM, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)				
IT	Fluoropolymers, uses RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (acrylic, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)				
IT	Epichlorohydrin rubber Synthetic rubber, uses RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (allyl glycidyl ether-epichlorohydrin-ethylene oxide, Epichlormer CG, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)				
IT	Epoxy resins, uses Polyureas Thermoplastic rubber RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses) (binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for				

electrophotog. apparatus)

IT **Surfactants**  
(conductive polymers; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Films**  
(elec. conductive; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **EPDM rubber**  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(ethylene-ethylideneborbornene-propene, Esprene 505, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Electric conductors**  
(films; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Acrylic polymers, uses**  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(fluorine-containing, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Nitrile rubber, uses**  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(hydrogenated, Zetpol 0020, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Polyimides, uses**  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(polyamide-, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Polyamides, uses**  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(polyimide-, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Conducting polymers**  
(polypyrroles, sulfonic group-containing; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Conducting polymers**  
(polythiophenes, sulfonic group-containing; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Electrophotographic apparatus**  
(rollers; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT **Conducting polymers**  
**Semiconductor materials**  
(semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog.

apparatus)

IT Polymer blends  
RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
(semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT Polyanilines  
RL: DEV (Device component use); IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(sulfonic group-containing, conductive polymer; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 220669-44-3P  
RL: DEV (Device component use); IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 9011-14-7, LG 6A 434322-68-6, Defensa TR 230K 577796-28-2, Vylomax HR 16NN  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 9003-17-2  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(butadiene rubber, JSR-BR 1220NM, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 62-53-3DP, Aniline, polymers with sulfonic acid group-containing surfactant 109-97-7DP, Pyrrole, polymers with dinonylnaphthalenesulfonic acid 110-02-1DP, Thiophene, polymers with dinonylnaphthalenesulfonic acid 22582-76-9DP, 9-Anthracenesulfonic acid, polymers with aniline 25322-17-2DP, Dinonylnaphthalenesulfonic acid, polymers with aniline 189376-87-2DP, 2,2'-Dinaphthylmethane-6,6'-disulfonic acid monosodium salt, polymers with aniline 712272-86-1DP, polymers with aniline  
RL: DEV (Device component use); IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(conductive polymer; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 175834-23-8, Burnock DB-980K  
RL: DEV (Device component use); RCT (Reactant); TEM (Technical or engineered material use); RACT (Reactant or reagent); USES (Uses)  
(crosslinking agent; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 9003-18-3  
RL: DEV (Device component use); POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(nitrile rubber, hydrogenated, Zetpol 0020, binder; semiconductor compns. containing conductive surfactant polymers with good heat, moisture, and voltage resistance for electrophotog. apparatus)

IT 11109-50-5, SUS 304

RL: DEV (Device component use); USES (Uses)  
 (roll core; semiconductor compns. containing **conductive**  
 surfactant polymers with good heat, moisture, and voltage resistance  
 for electrophotog. apparatus)

L32 ANSWER 6 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2004:485827 HCAPLUS

DN 141:24577

TI **Conductive composition of a fluorinated**  
**polymer which contains polyaniline, manufacturing**  
**process, and conductive films**

IN Korzhenko, Alexander; Pud, Alexander; Shapoval, Galina

PA Atofina, Fr.

SO Eur. Pat. Appl., 11 pp.

CODEN: EPXXDW

DT Patent

LA English

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 1428857	A1	20040616	EP 2002-293103	20021213
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO, MK, CY, AL, TR, BG, CZ, EE, SK				
	JP 2004197095	A2	20040715	JP 2003-417111	20031215
	US 2004181011	A1	20040916	US 2003-734596	20031215
PRAI	EP 2002-293103	A	20021213		
	US 2002-435256P	P	20021223		

OS MARPAT 141:24577

AB The process involves mixing (a) an aqueous solution of an anilinium salt with an aqueous dispersion of a fluorinated polymer, (b) then an oxidant is added to the mixture of step (a) to make a blend of the fluorinated polymer and doped polyaniline (PANI), (c) byproducts and unreacted aniline are removed by washing with H<sub>2</sub>O or an alc. to get a blend of purified fluorinated polymer and doped PANI, (d) eventually the purified fluorinated polymer and doped PANI of step (c) can be mixed with an acid, (e) H<sub>2</sub>O is removed from the purified fluorinated polymer and doped PANI of step (c) or (d) if any and the remaining powder is melted and shaped into films, pellets or any object. The ultimate dry powder green composite produced from 2 wt% PANI x p-toluenesulfonic acid and 98 wt% PVDF is used for melt compression at 180° to produce a dark green film with conductivity 8 + 10-8 S/cm.

IC ICM C08L079-02

ICS C08L027-12; C08L027-16; C08L027-18; C08L027-20; H01B001-12

CC 37-6 (Plastics Manufacture and Processing)

ST polyaniline tosylate blend PVDF conductive film

IT Fluoropolymers, properties

RL: POF (Polymer in formulation); PRP (Properties); USES (Uses)  
 (Kynar 9000; conductive composite of a fluorinated polymer which contains doped polyaniline for melt processable powder)

IT Polyanilines

RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP (Properties); PREP (Preparation); USES (Uses)  
 (conductive composite of a fluorinated polymer which contains doped polyaniline for melt

*applicant*

processable powder)

IT Fluoropolymers, properties  
 RL: POF (Polymer in formulation); PRP (Properties); USES (Uses)  
 (conductive composite of a fluorinated  
 polymer which contains doped polyaniline for melt  
 processable powder)

IT Films  
 (elec. conductive; conductive composite of a  
 fluorinated polymer which contains doped  
 polyaniline for melt processable powder)

IT Electric conductors  
 (films; conductive composite of a  
 fluorinated polymer which contains doped  
 polyaniline for melt processable powder)

IT 24937-79-9, Vinylidene fluoride homopolymer  
 RL: POF (Polymer in formulation); PRP (Properties); USES (Uses)  
 (Kynar 9000; conductive composite of a fluorinated  
 polymer which contains doped polyaniline for melt  
 processable powder)

IT 25233-30-1P, Polyaniline 147988-10-1P  
 193701-93-8P  
 RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP  
 (Properties); PREP (Preparation); USES (Uses)  
 (conductive composite of a fluorinated  
 polymer which contains doped polyaniline for melt  
 processable powder)

IT 24937-79-9, Vinylidene fluoride homopolymer  
 RL: POF (Polymer in formulation); PRP (Properties); USES (Uses)  
 (Kynar 9000; conductive composite of a fluorinated  
 polymer which contains doped polyaniline for melt  
 processable powder)

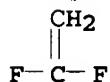
RN 24937-79-9 HCAPLUS

CN Ethene, 1,1-difluoro-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 75-38-7

CMF C2 H2 F2



IT 147988-10-1P 193701-93-8P  
 RL: IMF (Industrial manufacture); POF (Polymer in formulation); PRP  
 (Properties); PREP (Preparation); USES (Uses)  
 (conductive composite of a fluorinated  
 polymer which contains doped polyaniline for melt  
 processable powder)

RN 147988-10-1 HCAPLUS

CN Benzenamine, 4-methylbenzenesulfonate, homopolymer (9CI) (CA INDEX NAME)

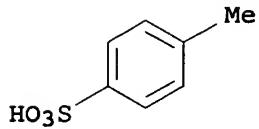
CM 1

CRN 14034-57-2

CMF C7 H8 O3 S . C6 H7 N

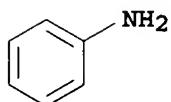
CM 2

CRN 104-15-4  
 CMF C7 H8 O3 S



CM 3

CRN 62-53-3  
 CMF C6 H7 N



RN 193701-93-8 HCAPLUS

CN Benzenesulfonic acid, dodecyl-, compd. with benzenamine (1:1), homopolymer  
 (9CI) (CA INDEX NAME)

CM 1

CRN 52193-59-6  
 CMF C18 H30 O3 S . C6 H7 N

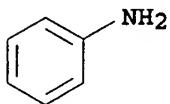
CM 2

CRN 27176-87-0  
 CMF C18 H30 O3 S  
 CCI IDS

D1- SO<sub>3</sub>HMe- (CH<sub>2</sub>)<sub>11</sub>- D1

CM 3

CRN 62-53-3  
 CMF C6 H7 N



RE.CNT 2 THERE ARE 2 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 7 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2004:431502 HCAPLUS  
 DN 141:126244  
 TI Proton Conductive Polyimide Electrolytes Containing Fluorenyl Groups: Synthesis, Properties, and Branching Effect  
 AU Miyatake, Kenji; Zhou, Hua; Watanabe, Masahiro  
 CS Clean Energy Research Center, University of Yamanashi, Kofu, 400-8510, Japan  
 SO Macromolecules (2004), 37(13), 4956-4960  
 CODEN: MAMOBX; ISSN: 0024-9297  
 PB American Chemical Society  
 DT Journal  
 LA English  
 AB Novel sulfonated polyimide copolymers as electrolytes for high-temperature fuel cell applications are reported. Sulfonated polyimide copolymers (SPIH-X; X refers to molar percentage of fluorenyl content) containing 0-60 mol % of fluorenyl groups as hydrophobic component were synthesized, of which electrolyte properties were studied and compared to those of the perfluorinated ionomer (Nafion 112). High-mol.-weight copolymers with good film-forming capability were obtained. Thermal stability with decomposition temperature of ca. 280 °C and no glass transition temperature was confirmed for the copolymers. SPIH shows unique water uptake behavior with the maximum value of 57% at X = 30. Water mols. absorbed in the electrolyte membrane with this specific composition do not evaporate easily so that the high proton conductivity of 1.67 S cm<sup>-1</sup> was obtained at 120° and 100% RH. The branching and crosslinking of SPIH-30 were carried out by applying 2 mol % of trifunctional monomer (melamine) in the polymerization and by electron beam irradiation upon the membrane. The branching and crosslinking are effective to improve oxidative stability and mech. strength. Although the proton conductivity decreases slightly by the branching and crosslinking, it still remains at the comparable level to that of Nafion 112.  
 CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
 Section cross-reference(s): 35, 38, 72, 76  
 ST proton conductive polyimide electrolyte fluorenyl group  
 branching sulfonated membrane; fuel cell separator membrane  
 polyelectrolyte arom polyimide mech strength  
 IT Polyimides, uses  
 RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (fluorene group- and fluorine-containing, cardo, aryl, sulfonate-containing; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)  
 IT Polyoxyalkylenes, uses  
 RL: DEV (Device component use); USES (Uses)  
 (fluorine- and sulfo-containing, ionomers, electrode binder; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)  
 IT Current density

(from methanol crossover, voltage and humidity effect on; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Electric current-potential relationship  
(of assembled fuel cell; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Stability  
(oxidative; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Carbon fibers, uses  
RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
(paper, anode support; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Fluoropolymers, uses  
RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
(polyimide-, fluorene group-containing, cardo, aryl, sulfonate-containing; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Cardo polymers  
RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
(polyimides, fluorene group- and fluorine-containing, aryl, sulfonate-containing; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Fluoropolymers, uses  
RL: DEV (Device component use); USES (Uses)  
(polyoxyalkylene-, sulfo-containing, ionomers, electrode binder; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Ionomers  
RL: DEV (Device component use); USES (Uses)  
(polyoxyalkylenes, fluorine- and sulfo-containing, electrode binder; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Ion exchange membranes  
(preparation and ion exchange capacity of; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Ionic conductivity  
(proton; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Crosslinking  
(radiochem.; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Fuel cell separators  
Fuel cells  
Membrane electrodes  
Membranes, nonbiological  
Polyelectrolytes  
(synthesis, properties, and DMFC performance of proton

conductive polyimide electrolytes containing trifluoromethyl groups)

IT Carbon black, uses  
 RL: DEV (Device component use); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT Fluoropolymers, uses  
 RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 117-61-3P, 4,4'-Diamino-2,2'-biphenyldisulfonic acid  
 RL: PUR (Purification or recovery); RCT (Reactant); PREP (Preparation); RACT (Reactant or reagent)  
 (DAPS; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 15499-84-0, 4,4'-(9-Fluorenylidene)dianiline  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (FDA; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 7732-18-5, Water, processes  
 RL: PEP (Physical, engineering or chemical process); PRP (Properties); PYP (Physical process); PROC (Process)  
 (absorption of; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 7720-78-7, Ferrous sulfate  
 RL: CAT (Catalyst use); USES (Uses)  
 (for oxidative stability; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 500783-35-7P  
 RL: PRP (Properties); PUR (Purification or recovery); SPN (Synthetic preparation); PREP (Preparation)  
 (plain and crosslinked; synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 42615-02-1  
 RL: CAT (Catalyst use); DEV (Device component use); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 67-56-1, Methanol, uses  
 RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); TEM (Technical or engineered material use); PROC (Process); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 7440-06-4, Platinum, uses 7440-57-5, Gold, uses  
 RL: DEV (Device component use); TEM (Technical or engineered material use); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

IT 108-39-4, m-Cresol, uses

RL: NUU (Other use, unclassified); USES (Uses)  
 (synthesis, properties, and DMFC performance of proton  
 conductive polyimide electrolytes containing trifluoromethyl  
 groups)

IT 481001-37-0P 724457-95-8P

RL: PRP (Properties); PUR (Purification or recovery); SPN (Synthetic  
 preparation); PREP (Preparation)  
 (synthesis, properties, and DMFC performance of proton  
 conductive polyimide electrolytes containing trifluoromethyl  
 groups)

IT 163294-14-2, Nafion 112

RL: PRP (Properties); TEM (Technical or engineered material use); USES  
 (Uses)  
 (synthesis, properties, and DMFC performance of proton  
 conductive polyimide electrolytes containing trifluoromethyl  
 groups)

IT 65-85-0, Benzoic acid, reactions 81-30-1, 1,4,5,8-  
 Naphthalenetetracarboxylic dianhydride 108-78-1, Melamine, reactions  
 121-44-8, Triethylamine, reactions 7722-84-1, Hydrogen peroxide,  
 reactions

RL: RCT (Reactant); RACT (Reactant or reagent)  
 (synthesis, properties, and DMFC performance of proton  
 conductive polyimide electrolytes containing trifluoromethyl  
 groups)

IT 500783-35-7P

RL: PRP (Properties); PUR (Purification or recovery); SPN (Synthetic  
 preparation); PREP (Preparation)  
 (plain and crosslinked; synthesis, properties, and DMFC performance of  
 proton conductive polyimide electrolytes containing  
 trifluoromethyl groups)

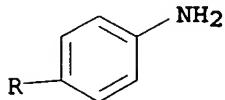
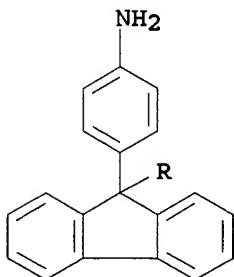
RN 500783-35-7 HCAPLUS

CN [1,1'-Biphenyl]-2,2'-disulfonic acid, 4,4'-diamino-, compd. with  
 N,N-diethylethanamine (1:2), polymer with [2]benzopyrano[6,5,4-  
 def] [2]benzopyran-1,3,6,8-tetrone and 4,4'-(9H-fluoren-9-  
 ylidene)bis[benzenamine] (9CI) (CA INDEX NAME)

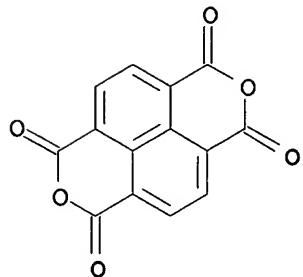
CM 1

CRN 15499-84-0

CMF C25 H20 N2



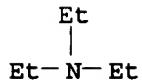
CM 2

CRN 81-30-1  
CMF C14 H4 O6

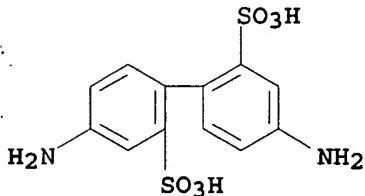
CM 3

CRN 481001-36-9  
CMF C12 H12 N2 O6 S2 . 2 C6 H15 N

CM 4

CRN 121-44-8  
CMF C6 H15 N

CM 5

CRN 117-61-3  
CMF C12 H12 N2 O6 S2

IT 481001-37-0P 724457-95-8P

RL: PRP (Properties); PUR (Purification or recovery); SPN (Synthetic preparation); PREP (Preparation)  
 (synthesis, properties, and DMFC performance of proton conductive polyimide electrolytes containing trifluoromethyl groups)

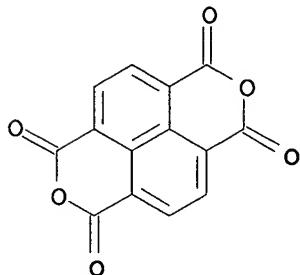
RN 481001-37-0 HCPLUS

CN [1,1'-Biphenyl]-2,2'-disulfonic acid, 4,4'-diamino-, compd. with N,N-diethylethanamine (1:2), polymer with [2]benzopyrano[6,5,4-def] [2]benzopyran-1,3,6,8-tetrone (9CI) (CA INDEX NAME)

CM 1

CRN 81-30-1

CMF C14 H4 O6



CM 2

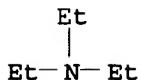
CRN 481001-36-9

CMF C12 H12 N2 O6 S2 . 2 C6 H15 N

CM 3

CRN 121-44-8

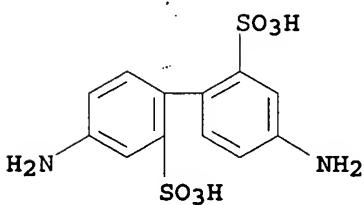
CMF C6 H15 N



CM 4

CRN 117-61-3

CMF C12 H12 N2 O6 S2

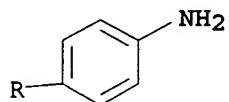
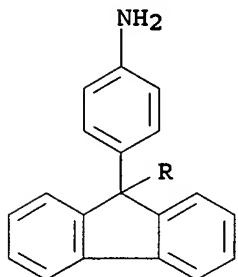


RN 724457-95-8 HCAPLUS

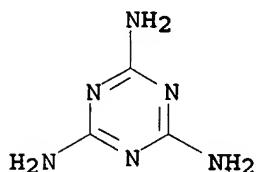
CN [1,1'-Biphenyl]-2,2'-disulfonic acid, 4,4'-diamino-, compd. with N,N-diethylethanamine (1:2), polymer with [2]benzopyrano[6,5,4-def] [2]benzopyran-1,3,6,8-tetrone, 4,4'-(9H-fluoren-9-ylidene)bis[benzenamine] and 1,3,5-triazine-2,4,6-triamine (9CI) (CA INDEX NAME)

INDEX NAME)

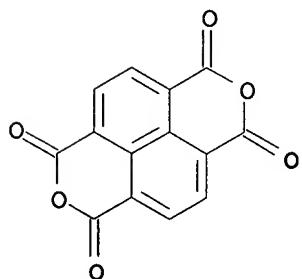
CM 1

CRN 15499-84-0  
CMF C25 H20 N2

CM 2

CRN 108-78-1  
CMF C3 H6 N6

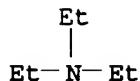
CM 3

CRN 81-30-1  
CMF C14 H4 O6

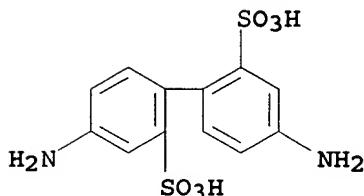
CM 4

CRN 481001-36-9  
CMF C12 H12 N2 O6 S2 . 2 C6 H15 N

CM 5

CRN 121-44-8  
CMF C6 H15 N

CM 6

CRN 117-61-3  
CMF C12 H12 N2 O6 S2RE.CNT 26 THERE ARE 26 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

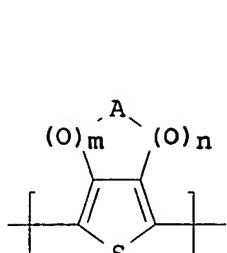
L32 ANSWER 8 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
 AN 2004:271590 HCAPLUS  
 DN 140:311998  
 TI Antistatic agent for antistatic film covering chemically amplified resist film, pattern formation using the antistatic film, and its use

IN Saita, Yoshihiro; Abe, Shinyoku  
 PA Showa Denko K. K., Japan  
 SO Jpn. Kokai Tokkyo Koho, 20 pp.  
 CODEN: JKXXAF

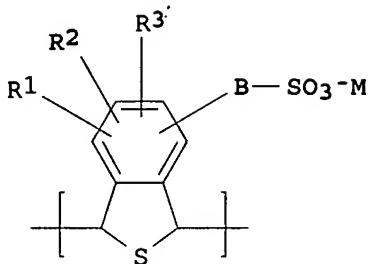
DT Patent  
 LA Japanese

FAN.CNT 1

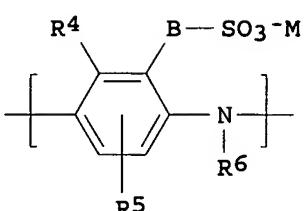
PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2004099678	A2	20040402	JP 2002-260957	20020906
PRAI JP 2002-260957		20020906		
OS MARPAT 140:311998				
GI				



I



II



III

AB The antistatic agent with good pH stability contains water-soluble elec. conductive polymers, fluorinated aliphatic amines, and water. Preferably, the amines comprise  $\geq 1$  represented by the general formula  $X(CF_2)_mCY_2NH_2$  ( $X = F, OH; Y = H, F; m = 1-10$  integer), more preferably, trifluoroethylamine. Preferably, the water-soluble elec. conductive polymers comprise  $\pi$ -conjugated ones bearing Broensted acid groups, more preferably, sulfonic acid groups. More preferably, the water-soluble elec. conductive polymers are represented by general formulas I [ $m, n = 0, 1; A = C1-4 alkylene, C1-4 alkenylene$  which have  $\geq 1$  BSO<sub>3</sub>-M, may be substituted with halo, OH, NO<sub>2</sub>, etc., and may contain  $\geq 2$  C:C; B = (CH<sub>2</sub>)<sub>p</sub>[O(CH<sub>2</sub>)<sub>q</sub>]<sub>r</sub>; p = 0-5 integer, q = 1-3 integer, r = 0-3 integer; M = H<sup>+</sup>, alkali metal ion, quaternary ammonium ion], II or III [R<sub>1</sub>-R<sub>5</sub> = H, C<sub>1-20</sub> hydrocarbyl, alkoxy, alkylester, OH, halo, NO<sub>2</sub>, BSO<sub>3</sub>M, etc.; alkyl, alkoxy, or alkylester groups of R<sub>1</sub>-R<sub>5</sub> may contain CO, ether, CO<sub>2</sub>, SO<sub>3</sub>, amido, sulfoneamido, sulfide, S(O), SO<sub>2</sub>, :NH, thioether in the chain; R<sub>6</sub> = H, C<sub>1-20</sub> hydrocarbyl or Ph which may be substituted; B, p, q, r, M = same as I]. The water-soluble elec. conductive polymers may contain 5-sulfoisothianaphthene-1,3-diyl as the chemical structure. The antistatic treatment agent may contain surfactants. Chemical-amplified resist films are covered with films of the antistatic treatment agent. Semiconductor elements, photomasks, reticles, glass substrates, quartz substrates, GMR heads, or magnetic substrates are fabricated by using the antistatic treatment agent.

IC ICM C09K003-16  
ICS C08G061-12; C08L079-00; G03F007-11; H01L021-027

CC 74-5 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

ST Section cross-reference(s): 38, 76, 77  
trifluoroethylamine neutralized polysulfoisothianaphthene diyl antistatic agent; chem amplified film coating antistatic agent; fluorinated aliphatic amine neutralizing agent; self doped elec conductive polymer neutralizing

IT Magnetic recording heads  
(GMR; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use in fabrication of)

IT Amines, uses  
RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)  
(aliphatic, fluorinated; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT Antistatic agents  
Electron beam resists  
(buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT Glass substrates  
Photomasks (lithographic masks)  
Semiconductor devices  
(buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use in fabrication of)

IT Giant magnetoresistance  
(heads; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use in fabrication of)

IT Conducting polymers  
(polythiophenes; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT Magnetic materials  
(substrates; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use in fabrication of)

IT Conducting polymers  
(water-soluble; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT 119574-53-7, SAL 601  
RL: TEM (Technical or engineered material use); USES (Uses)  
(EB resist; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT 27176-87-0, Dodecylbenzenesulfonic acid 29010-16-0, Trifluoroethylamine  
RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)  
(buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT 188754-53-2, Poly(5-sulfoisothianaphthene-1,3-diyl)  
RL: TEM (Technical or engineered material use); USES (Uses)  
(buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT 247072-90-8, NEB 22  
RL: TEM (Technical or engineered material use); USES (Uses)

(neg. EB resist; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

IT 14808-60-7, Quartz, uses  
RL: TEM (Technical or engineered material use); USES (Uses)  
(substrates; buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use in fabrication of)

IT 25233-30-1D, Polyaniline, sulfonated  
RL: TEM (Technical or engineered material use); USES (Uses)  
(Buffered water-soluble elec. conductive polymer-based antistatic agent for antistatic coating on chemical amplified resist film, its patterning, and its use)

L32 ANSWER 9 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
AN 2002:872173 HCAPLUS

DN 138:228353

TI Electropolymerization and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H

AU Sahin, Yucel; Pekmez, Kadir; Yildiz, Attila

CS Faculty of Science, Department of Chemistry, Anadolu University, Eskisehir, 26470, Turk.

SO Synthetic Metals (2002), 131(1-3), 7-14

CODEN: SYMEDZ; ISSN: 0379-6779

PB Elsevier Science B.V.

DT Journal

LA English

AB In situ sulfonation reaction of aniline was carried out in water and water-acetonitrile mixture by changing both aniline and fluorosulfonic acid (FSO<sub>3</sub>H) concns. The optimum conditions for the polymer formation was determined. The dry conductivity values of the film increased and the solubilities of the polymer decreased with increasing volume of acetonitrile in the mixture. The polymer film was found to grow much faster compared to the growth in non-aqueous acetonitrile. FTIR, elemental anal., cyclic voltammetry and UV-VIS spectroscopic methods were used to characterize the polymers.

CC 72-9 (Electrochemistry)

Section cross-reference(s): 38, 73

ST aniline electropolymer sulfonation water acetonitrile fluorosulfonic acid

IT Polymerization

(electrochem.; and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H)

IT Sulfonation

(electropolymer. and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H)

IT Cyclic voltammetry

(of Pt electrode in water-acetonitrile mixture containing FSO<sub>3</sub>H in presence of aniline and formed polyaniline film in FSO<sub>3</sub>H solution containing pyridine)

IT Optimization

(of conditions for electropolymer. and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H)

IT UV and visible spectra

(of polyaniline film formed on Pt electrode in presence of FSO<sub>3</sub>H in water-acetonitrile and in DMSO)

IT IR spectra

(of polyaniline film formed on Pt electrode in water-acetonitrile solution containing FSO<sub>3</sub>H)

IT Oxidation, electrochemical  
(of polyaniline film on Pt electrode in FSO<sub>3</sub>H solution containing pyridine)

IT Solubility  
(of polyaniline film on Pt electrode in basic aqueous solns., DMSO and NMP)

IT 110-86-1, Pyridine, reactions  
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent) (cyclic voltammetry of Pt electrode coated with polyaniline film in FSO<sub>3</sub>H solution containing)

IT 62-53-3, Aniline, reactions  
RL: CPS (Chemical process); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent) (electropolymer. and in situ sulfonation in water-acetonitrile mixture containing FSO<sub>3</sub>H)

IT 7789-21-1, Fluorosulfonic acid  
RL: CPS (Chemical process); NUU (Other use, unclassified); PEP (Physical, engineering or chemical process); RCT (Reactant); PROC (Process); RACT (Reactant or reagent); USES (Uses) (electropolymer. and in situ sulfonation of aniline in water-acetonitrile mixture containing)

IT 75-05-8, Acetonitrile, uses 7732-18-5, Water, uses  
RL: NUU (Other use, unclassified); USES (Uses) (electropolymer. and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H)

IT 7440-06-4, Platinum, uses  
RL: DEV (Device component use); USES (Uses) (electropolymer. and in situ sulfonation of aniline in water-acetonitrile mixture containing FSO<sub>3</sub>H on Pt electrode)

IT 67-68-5, DMSO, uses 872-50-4, NMP, uses  
RL: NUU (Other use, unclassified); USES (Uses) (of polyaniline film formed on Pt electrode in)

IT 1310-58-3, Potassium hydroxide, uses  
RL: NUU (Other use, unclassified); USES (Uses) (of polyaniline film formed on Pt electrode in solution of)

RE.CNT 33 THERE ARE 33 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 10 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 2002:439119 HCAPLUS

DN 137:21065

TI Electrically conductive polymer compositions with uniform resistivity and high voltage-resistance dependence

IN Yoshikawa, Hitoshi; Suzuki, Satoshi; Ito, Kunio

PA Tokai Rubber Industries, Ltd., Japan

SO Jpn. Kokai Tokkyo Koho, 9 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 2002167519	A2	20020611	JP 2001-252690	20010823
	JP 2006089755	A2	20060406	JP 2005-350302	20051205
	JP 2006124717	A2	20060518	JP 2005-350301	20051205
	JP 2006146249	A2	20060608	JP 2005-350300	20051205
PRAI	JP 2000-284068	A	20000919		
	JP 2001-252690	A3	20010823		

AB The compns., useful for electrophotog. components, contain (A) elec. conductive polymers having surfactant structures and (B) binder polymers. Thus, 10 parts aniline and 8 parts dodecylbenzenesulfonic acid were oxidation-polymerized to give an elec. conductive polymer, which was blended with 82 parts poly(Me methacrylate) and extruded on a glass plate to give a 100- $\mu$ m elec. conductive film with elec. resistivity 1.5 + 106  $\Omega$ -cm and high resistivity dependence on temperature, humidity, and voltage.

IC ICM C08L101-12  
ICS C08K003-00; C08L101-00; F16C013-00; G03G015-00; G03G015-02; G03G015-08; G03G015-16; G03G021-06; G03G021-10

CC 37-6 (Plastics Manufacture and Processing)  
Section cross-reference(s): 42, 76

ST conductive surfactant polymer blend voltage resistance dependence; aniline dodecylbenzenesulfonic acid copolymer PMMA blend conductor coating

IT Carbon black, uses  
RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)  
(Denka Black HS 100, conducting agent; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Fluoropolymers, uses  
RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(acrylic, binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Acrylic polymers, uses  
Epoxy resins, uses  
Polyamides, uses  
Polyurethanes, uses  
RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)  
(binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Conducting polymers  
(elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Polyanilines  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
(elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Polymer blends  
RL: TEM (Technical or engineered material use); USES (Uses)  
(elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Coating materials  
(elec. conductive; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Acrylic polymers, uses  
RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(fluorine-containing, binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT Nitrile rubber, uses

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(hydrogenated, Zetpol 0020, binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT 9011-14-7, Poly(methyl methacrylate) 85510-39-0, EF 30T 434322-68-6, Defensa TR 230K

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT 32503-27-8, Tetrabutylammonium hydrogensulfate

RL: MOA (Modifier or additive use); TEM (Technical or engineered material use); USES (Uses)

(conducting agent; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT 132512-01-7P, Aniline-dodecylbenzenesulfonic acid copolymer

433731-72-7P 433731-73-8P

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT 9003-18-3

RL: POF (Polymer in formulation); TEM (Technical or engineered material use); USES (Uses)

(nitrile rubber, hydrogenated, Zetpol 0020, binder; elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

IT 132512-01-7P, Aniline-dodecylbenzenesulfonic acid copolymer

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(elec. conductive polymer compns. containing conductive polymers having surfactant structure with uniform resistivity and high voltage-resistance dependence)

RN 132512-01-7 HCPLUS

CN Benzenesulfonic acid, dodecyl-, polymer with benzenamine (9CI) (CA INDEX NAME)

CM 1

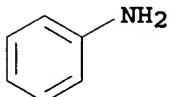
CRN 27176-87-0

CMF C18 H30 O3 S

CCI IDS

D1- SO<sub>3</sub>HMe- (CH<sub>2</sub>)<sub>11</sub>- D1

CM 2

CRN 62-53-3  
CMF C6 H7 N

L32 ANSWER 11 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN  
 AN 2001:403574 HCPLUS  
 DN 135:26829  
 TI Thermally stable semiconductive polyamic acid composition and its application in electrophotographic copying machine  
 IN Nishiura, Naoki  
 PA Gunze, Ltd., Japan  
 SO Jpn. Kokai Tokkyo Koho, 10 pp.  
 CODEN: JKXXAF  
 DT Patent  
 LA Japanese  
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2001152013	A2	20010605	JP 1999-335764	19991126
PRAI JP 1999-335764		19991126		
AB	The composition contains 10-25% carbon black with a structure index, i.e., [DBP (di-Bu phthalate) absorption; g/100 g]/(volatile content; weight%), <100 and 75-90% polyamic acid. The composition is rotationally molded without centrifugal force to give a thermally stable endless tubular polyimide film. The tubular polyimide film is preferably coated with a silicone rubber, fluoro rubber, or fluorosilicone rubber layer and has elec. conductivity $\leq 1 \mu\text{S}/\text{cm}^2$ and electrostatic capacitance 10-400 pF/cm <sup>2</sup> . The tubular film is used as a belt for intermediate image transfer or for heat fixing of image in a color electrophotog. copying machine.			
IC	ICM C08L079-08 ICS C08L079-08; B29C041-04; C08G073-10; C08J005-18; C08J007-04; C08K003-04; G03G015-16; G03G015-20; H01B001-24; B29K079-00; B29L023-00			
CC	74-3 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)			

ST Section cross-reference(s): 37, 38, 39, 76

ST thermally stable semiconductive polyamic acid compn; carbon black oil absorption volatile content; rotationally molded endless tubular film polyimide; silicone rubber polyimide tubular film belt; fluoro rubber polyimide tubular film belt; fluorosilicone rubber polyimide tubular film belt; color electrophotog coping machine belt polyimide

IT Fluoro rubber

RL: DEV (Device component use); USES (Uses)  
(GLS 213; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

IT Belts

(endless; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Silicone rubber, uses

RL: DEV (Device component use); USES (Uses)  
(fluorine-containing; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

IT Molding of plastics and rubbers

(rotational; of thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Fluoro rubber

RL: DEV (Device component use); USES (Uses)  
(silicone; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

IT Silicone rubber, uses

RL: DEV (Device component use); USES (Uses)  
(thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

IT Electric conductors

Electrophotographic apparatus

Heat-resistant materials

Resistors

(thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Polyamic acids

RL: DEV (Device component use); PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); PROC (Process); USES (Uses)  
(thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Polyimides, properties

RL: DEV (Device component use); POF (Polymer in formulation); PRP (Properties); USES (Uses)  
(thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Carbon black, uses

RL: MOA (Modifier or additive use); USES (Uses)  
(thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT Plastic films

(tubular; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog.

copying machine)

IT 9011-17-0, Hexafluoropropylene-vinylidene fluoride copolymer  
 RL: DEV (Device component use); USES (Uses)  
 (rubber; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

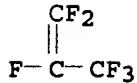
IT 9043-05-4P 25036-53-7P, 4,4'-Diaminodiphenyl ether-pyromellitic anhydride copolymer, sru 25038-81-7P, 4,4'-Diaminodiphenyl ether-pyromellitic anhydride copolymer 26875-02-5P, 3,3',4,4'-Benzophenonetetracarboxylic acid-4,4'-diaminodiphenylmethane copolymer 26913-87-1P 56802-71-2P  
 RL: DEV (Device component use); IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); PREP (Preparation); PROC (Process); USES (Uses)  
 (thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

IT 9011-17-0, Hexafluoropropylene-vinylidene fluoride copolymer  
 RL: DEV (Device component use); USES (Uses)  
 (rubber; thermally stable polyamic acid composition containing carbon black for endless tubular film for belt covered with)

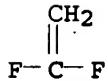
RN 9011-17-0 HCPLUS

CN 1-Propene, 1,1,2,3,3,3-hexafluoro-, polymer with 1,1-difluoroethylene (9CI)  
 (CA INDEX NAME)

CM 1

CRN 116-15-4  
 CMF C3 F6

CM 2

CRN 75-38-7  
 CMF C2 H2 F2

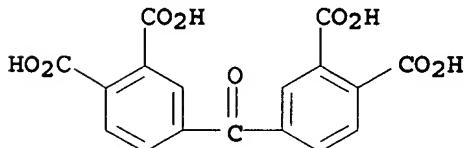
IT 26875-02-5P, 3,3',4,4'-Benzophenonetetracarboxylic acid-4,4'-diaminodiphenylmethane copolymer  
 RL: DEV (Device component use); IMF (Industrial manufacture); PEP (Physical, engineering or chemical process); POF (Polymer in formulation); PRP (Properties); PREP (Preparation); PROC (Process); USES (Uses)  
 (thermally stable polyamic acid composition containing carbon black for endless tubular film for belt in electrophotog. copying machine)

RN 26875-02-5 HCPLUS

CN 1,2-Benzenedicarboxylic acid, 4,4'-carbonylbis-, polymer with  
4,4'-methylenebis[benzenamine] (9CI) (CA INDEX NAME)

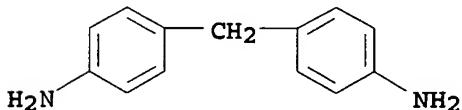
CM 1

CRN 2479-49-4  
CMF C17 H10 O9



CM 2

CRN 101-77-9  
CMF C13 H14 N2



L32 ANSWER 12 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 2001:109971 HCPLUS

DN 134:164180

TI Block polyimide thin films, their manufacture by casting, and  
their use

IN Matsumoto, Shunichi; Itatani, Hiroshi

PA PI Gijitsu Kenkyusho K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 8 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 2001040108	A2	20010213	JP 1999-246027	19990728
PRAI JP 1999-246027		19990728		

AB The thin films are manufactured by casting polyimide block copolymer solns. on a substrate film, followed by drying and peeling the copolymer off the substrate film. The films are used as (A) heat-resistant thin-film capacitors by lamination with elec. conductive thin films, (B) semiconductor passivation films by lamination on a semiconductor, and (C) an elec. insulator by applying on an elec. circuit substrate. Thus, a block polyimide solution [prepared from 3,4,3',4'-benzophenonetetracarboxylic dianhydride, 2,4-diaminotoluene, 3,4,3',4'-biphenyltetracarboxylic anhydride, and 2,2-bis[4-(4-aminophenoxy)phenyl]propane] was cast on a PET film, dried, and peeled off the PET film to give a 3-μm polymeric film.

IC ICM C08J005-18

ICS C08G073-10  
CC 38-3 (Plastics Fabrication and Uses)  
Section cross-reference(s): 76  
ST block polyimide cast film elec insulator; capacitor cast  
film block polyimide; semiconductor passivation film  
block polyimide  
IT Passivation  
(film for; manufacture of block polyimide thin films for  
capacitors, elec. insulators, and semiconductor passivation  
films)  
IT Polyimides, uses  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(fluorine-contg, block; manufacture of block polyimide thin films  
for capacitors, elec. insulators, and semiconductor passivation  
films)  
IT Capacitors  
Electric insulators  
(manufacture of block polyimide thin films for capacitors, elec.  
insulators, and semiconductor passivation films)  
IT Polyimides, uses  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyether-, block; manufacture of block polyimide thin films for  
capacitors, elec. insulators, and semiconductor passivation  
films)  
IT Polyimides, uses  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyether-, fluorine-containing, block; manufacture of block polyimide thin  
films for capacitors, elec. insulators, and semiconductor  
passivation films)  
IT Fluoropolymers, uses  
Polyketones  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyether-polyimide-, block; manufacture of block polyimide thin  
films for capacitors, elec. insulators, and semiconductor  
passivation films)  
IT Polyimides, uses  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyether-polyketone-, block; manufacture of block polyimide thin  
films for capacitors, elec. insulators, and semiconductor  
passivation films)  
IT Fluoropolymers, uses  
Polyethers, uses  
Polyketones  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyimide-, block; manufacture of block polyimide thin films for  
capacitors, elec. insulators, and semiconductor passivation  
films)  
IT Polyethers, uses  
RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or  
engineered material use); PREP (Preparation); USES (Uses)  
(polyimide-, fluorine-containing, block; manufacture of block polyimide thin  
films for capacitors, elec. insulators, and semiconductor  
passivation films)  
IT Polyethers, uses

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (polyimide-polyketone-, block; manufacture of block polyimide thin films for capacitors, elec. insulators, and semiconductor passivation films)

IT Polyimides, uses

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (polyketone-, block; manufacture of block polyimide thin films for capacitors, elec. insulators, and semiconductor passivation films)

IT 15499-84-0DP, 9,9-Bis(4-aminophenyl)fluorene, polyimide block polymers

69563-88-8DP, 2,2-Bis[4-(4-

aminophenoxy)phenyl]hexafluoropropane, polyimide block polymers  
 177190-45-3P 324750-74-5P 324750-77-8P 325467-77-4P

325467-78-5P 325467-79-6P 325467-80-9P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (manufacture of block polyimide thin films for capacitors, elec. insulators, and semiconductor passivation films)

IT 324750-74-5P 324750-77-8P 325467-79-6P

RL: IMF (Industrial manufacture); PRP (Properties); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(manufacture of block polyimide thin films for capacitors, elec. insulators, and semiconductor passivation films)

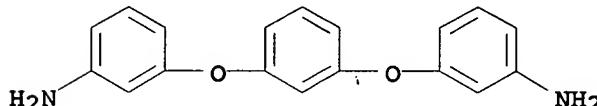
RN 324750-74-5 HCPLUS

CN Benzoic acid, 3,5-diamino-, polymer with [5,5'-biisobenzofuran]-1,1',3,3'-tetrone, 3a,4,4a,7a,8,8a-hexahydro-4,8-etheno-1H,3H-benzo[1,2-c:4,5-c']difuran-1,3,5,7-tetrone and 3,3'-[1,3-phenylenebis(oxy)]bis[benzenamine], block (9CI) (CA INDEX NAME)

CM 1

CRN 10526-07-5

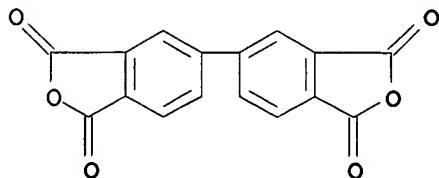
CMF C18 H16 N2 O2



CM 2

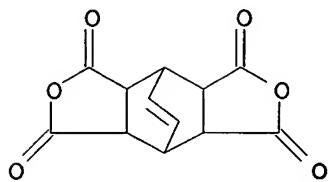
CRN 2420-87-3

CMF C16 H6 O6



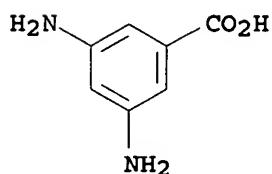
CM 3

CRN 1719-83-1  
 CMF C12 H8 O6



CM 4

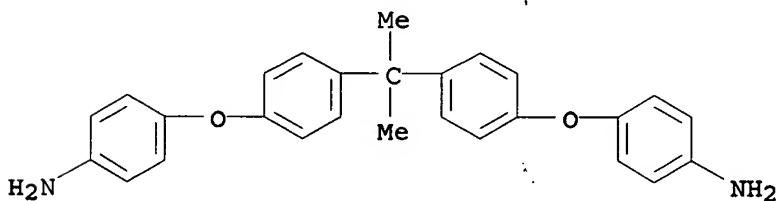
CRN 535-87-5  
 CMF C7 H8 N2 O2



RN 324750-77-8 HCAPLUS  
 CN Benzoic acid, 3,5-diamino-, polymer with [5,5'-biisobenzofuran]-1,1',3,3'-tetrone, 5,5'-carbonylbis[1,3-isobenzofurandione], 4-methyl-1,3-benzenediamine and 4,4'-[(1-methylethylidene)bis(4,1-phenyleneoxy)]bis[benzenamine], block (9CI) (CA INDEX NAME)

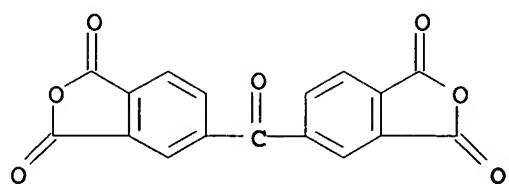
CM 1

CRN 13080-86-9  
 CMF C27 H26 N2 O2

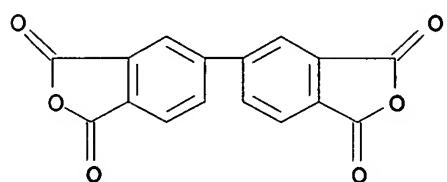


CM 2

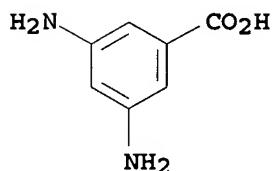
CRN 2421-28-5  
 CMF C17 H6 O7



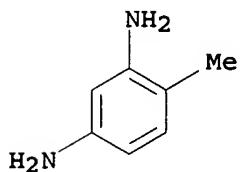
CM 3

CRN 2420-87-3  
CMF C16 H6 O6

CM 4

CRN 535-87-5  
CMF C7 H8 N2 O2

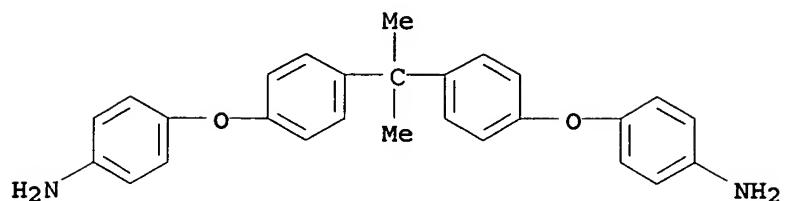
CM 5

CRN 95-80-7  
CMF C7 H10 N2

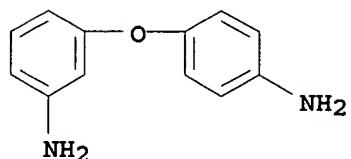
RN 325467-79-6 HCPLUS

CN Benzoic acid, 3,4-diamino-, polymer with 3-(4-aminophenoxy)benzenamine, [5,5'-biisobenzofuran]-1,1',3,3'-tetrone, 3a,4,4a,7a,8,8a-hexahydro-4,8-etheno-1H,3H-benzo[1,2-c:4,5-c']difuran-1,3,5,7-tetrone and 4,4'-(1-methylethylidene)bis(4,1-phenyleneoxy)]bis[benzenamine], block (9CI) (CA INDEX NAME)

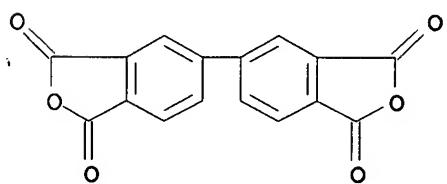
CM 1

CRN 13080-86-9  
CMF C27 H26 N2 O2

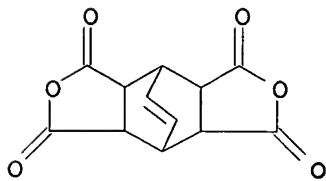
CM 2

CRN 2657-87-6  
CMF C12 H12 N2 O

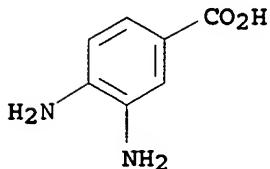
CM 3

CRN 2420-87-3  
CMF C16 H6 O6

CM 4

CRN 1719-83-1  
CMF C12 H8 O6

CM 5

CRN 619-05-6  
CMF C7 H8 N2 O2

L32 ANSWER 13 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN  
 AN 2000:570035 HCPLUS  
 DN 133:136539  
 TI Polyimide/fluoropolymer laminates, their fabrication, and insulating tape for wrapping around conductors  
 IN Nishinaka, Masaru; Ono, Kazuhiro; Akahori, Kiyokazu  
 PA Kaneka Corporation, Japan  
 SO Fr. Demande, 29 pp.

CODEN: FRXXBL

DT Patent  
LA French

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	FR 2786193	A1	20000526	FR 1999-14597	19991119
	FR 2786193	B1	20020322		
	JP 2000211081	A2	20000802	JP 1999-310944	19991101
	US 6475624	B1	20021105	US 1999-443382	19991119
PRAI	JP 1998-331728	A	19981120		

OS MARPAT 133:136539

AB The laminates are obtained by applying a fluoropolymer layer to one or both sides of a polyimide film which shows ≥80% retention of tear strength (ASTM D 1938) after 12 h at 150° and 100% relative humidity. Preferably the polyimide film contains a compound of Al, Si, Ti, Mn, Fe, Co, Cu, Zn, Sn, Sb, Pb, or Bi, incorporated at the polyamic acid stage before cyclization. Thus, 90 g of a 17% solution in DMF of a polyamic acid from pyromellitic anhydride 4, 4,4'-diaminodiphenyl ether 3, and p-phenylenediamine 1 mol was mixed with 0.1 g tributoxytitanium stearate, coated on an Al foil, gelled 2 min at 110°, separated from the foil, and heated 1 min each at 300°, 400°, and 500° to give a polyimide film 25 µm thick, which showed tear strength retention 87%, compared with 38% when the Ti compound was omitted. An aqueous dispersion of FEP was applied to both sides of the polyimide film at 2.5 µm each side, dried 1 min at 150°, and cured 15 s at 415° to give a laminated tape.

IC ICM C08J007-04

ICS C08L079-08; H01B003-30

CC 38-3 (Plastics Fabrication and Uses)

ST insulating tape polyimide fluoropolymer laminate

IT Polyimides, uses

Polyimides, uses

Polyimides, uses

RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM

(Technical or engineered material use); PREP (Preparation); USES (Uses)  
(polyester-polyether-; polyimide-fluoropolymer laminates as insulating  
tape for wrapping around conductors)

IT Polyethers, uses  
Polyethers, uses  
Polyethers, uses  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES (Uses)  
(polyester-polyimide-; polyimide-fluoropolymer laminates as insulating  
tape for wrapping around conductors)

IT Polyimides, uses  
Polyimides, uses  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES (Uses)  
(polyether-; polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT Polyesters, uses  
Polyesters, uses  
Polyesters, uses  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES (Uses)  
(polyether-polyimide-; polyimide-fluoropolymer laminates as insulating  
tape for wrapping around conductors)

IT Polyethers, uses  
Polyethers, uses  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES (Uses)  
(polyimide-; polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT Fluoropolymers, uses  
Laminated plastics, uses  
RL: TEM (Technical or engineered material use); USES (Uses)  
(polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT Electric insulators  
(tapes; polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT 25036-53-7P 25038-81-7P, 4,4'-Diaminodiphenyl ether-pyromellitic  
anhydride copolymer 31975-60-7P, 4,4'-Diaminodiphenyl  
ether-p-phenylenediamine-pyromellitic anhydride copolymer  
208934-81-0P  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES  
(Uses)  
(polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT 7772-99-8, Stannous chloride, uses 79110-90-0 81307-49-5  
RL: MOA (Modifier or additive use); USES (Uses)  
(polyimide-fluoropolymer laminates as insulating tape for wrapping  
around conductors)

IT 25067-11-2, Hexafluoropropylene-tetrafluoroethylene  
copolymer  
RL: TEM (Technical or engineered material use); USES (Uses)  
(polyimide-fluoropolymer laminates as insulating tape for  
wrapping around conductors)

IT 208934-81-0P  
RL: IMF (Industrial manufacture); POF (Polymer in formulation); TEM  
(Technical or engineered material use); PREP (Preparation); USES  
(Uses)  
(polyimide-fluoropolymer laminates as insulating tape for wrapping

around conductors)

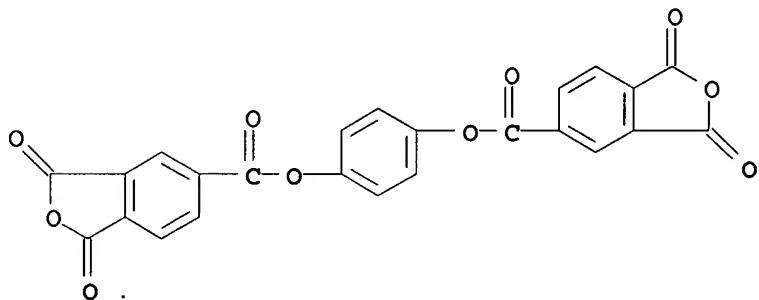
RN 208934-81-0 HCPLUS

CN 5-Isobenzofurancarboxylic acid, 1,3-dihydro-1,3-dioxo-, 1,4-phenylene ester, polymer with 1,4-benzenediamine, 1H,3H-benzo[1,2-c:4,5-c']difuran-1,3,5,7-tetrone and 4,4'-oxybis[benzenamine] (9CI) (CA INDEX NAME)

CM 1

CRN 2770-49-2

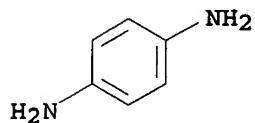
CMF C24 H10 O10



CM 2

CRN 106-50-3

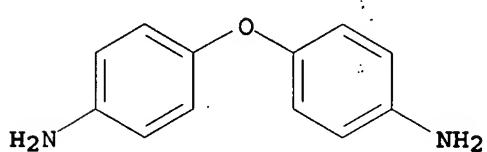
CMF C6 H8 N2



CM 3

CRN 101-80-4

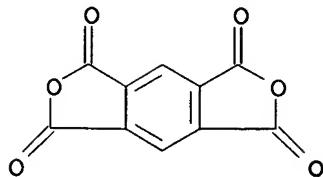
CMF C12 H12 N2 O



CM 4

CRN 89-32-7

CMF C10 H2 O6



IT 25067-11-2, Hexafluoropropylene-tetrafluoroethylene copolymer

RL: TEM (Technical or engineered material use); USES (Uses) (polyimide-fluoropolymer laminates as insulating tape for wrapping around conductors)

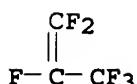
RN 25067-11-2 HCPLUS

CN 1-Propene, 1,1,2,3,3,3-hexafluoro-, polymer with tetrafluoroethylene (9CI) (CA INDEX NAME)

CM 1

CRN 116-15-4

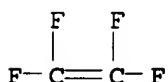
CMF C3 F6



CM 2

CRN 116-14-3

CMF C2 F4



L32 ANSWER 14 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 1999:575452 HCPLUS

DN 131:311132

TI Enhancement of electrical stability of polyaniline films in aqueous media by surface graft copolymerization with hydrophobic monomers

AU Zhao, Baozong; Neoh, K. G.; Liu, F. T.; Kang, E. T.; Tan, K. L.

CS Department of Chemical and Environmental Engineering, National University of Singapore, Kent Ridge, 119260, Singapore

SO Langmuir (1999), 15(23), 8259-8264

CODEN: LANGD5; ISSN: 0743-7463

PB American Chemical Society

DT Journal

LA English

AB Surface modification of free-standing polyaniline (PANI) films and PANI coating on low-d. polyethylene (LDPE) substrates via UV-induced graft copolymer. with hydrophobic monomers was carried out. Pentafluorostyrene (PFS) and styrene were successfully graft copolymerd. on the PANI surfaces, rendering them hydrophobic. The effects of UV graft

copolymn. time, graft copolymn.; temperature, and monomer concentration on the graft

concentration were investigated. The pristine and graft-modified films were characterized using both surface and bulk anal. techniques. For the pristine PANi films, the loss of counterions from the surface region of the film occurs very rapidly in deionized water. This loss is very effectively retarded by surface graft copolymn. with PFS, hence preserving the PANi's conductivity even upon prolonged immersion in deionized water. This enhancement in the elec. stability of the PANi film was also achieved in moderately basic aqueous medium.

CC 37-5 (Plastics Manufacture and Processing)

Section cross-reference(s): 76

ST polyaniline graft polymn elec stability enhancement

IT Contact angle

Hydrophobicity

(elec. stability enhancement and hydrophobization of polyaniline films in aqueous media by graft polymerization)

IT Conducting polymers

(elec. stability enhancement of polyaniline films in aqueous media by graft polymerization)

IT Polyanilines

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (elec. stability enhancement of polyaniline films in aqueous media by graft polymerization)

IT Fluoropolymers, preparation

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (elec. stability enhancement of polyaniline films in aqueous media by graft polymerization with fluorostyrene)

IT Polymerization

(graft, photochem.; elec. stability enhancement of polyaniline films in aqueous media by graft polymerization)

IT Polymerization

(graft, surface; elec. stability enhancement of polyaniline films in aqueous media by graft polymerization)

IT Polymer morphology

(surface; in elec. stability enhancement and hydrophobization of polyaniline films in aqueous media by graft polymerization)

IT 167762-99-4P, Aniline-styrene graft copolymer 247215-97-0P

RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation) (elec. stability enhancement of polyaniline films in aqueous media by graft polymerization)

RE.CNT 19 THERE ARE 19 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMAT

L32 ANSWER 15 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 1998:464360 HCPLUS

DN 129:122975

TI Salts of perfluorinated sulfonamides or sulfinamides and their use as ionic conductors and as catalysts

IN Armand, Michel; Choquette, Yves; Gauthier, Michel; Michot, Christophe  
PA Centre National de la Recherche Scientifique (CNRS), Fr.; Hydro-Quebec

SO Eur. Pat. Appl., 65 pp.

CODEN: EPXXDW

DT Patent

LA French

FAN.CNT 5

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	EP 850920	A2	19980701	EP 1997-403187	19971230
	EP 850920	A3	19980708		
	EP 850920	B1	20020911		
	R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
	CA 2194127	AA	19980630	CA 1996-2194127	19961230
	CA 2199231	AA	19980905	CA 1997-2199231	19970305
	CA 2244979	AA	19980709	CA 1997-2244979	19971230
	CA 2248242	AA	19980709	CA 1997-2248242	19971230
	CA 2248244	AA	19980709	CA 1997-2248244	19971230
	CA 2248246	AA	19980709	CA 1997-2248246	19971230
	CA 2248303	AA	19980709	CA 1997-2248303	19971230
	CA 2248304	AA	19980709	CA 1997-2248304	19971230
	WO 9829358	A2	19980709	WO 1997-CA1008	19971230
	WO 9829358	A3	19981008		
	W: CA, JP, US				
	RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
	WO 9829399	A1	19980709	WO 1997-CA1009	19971230
	W: CA, JP, US				
	WO 9829389	A1	19980709	WO 1997-CA1010	19971230
	W: CA, JP, US				
	WO 9829396	A1	19980709	WO 1997-CA1011	19971230
	W: CA, JP, US				
	WO 9829877	A1	19980709	WO 1997-CA1012	19971230
	W: CA, JP, US				
	RW: AT, BE, CH, DE, DK, ES, FI, FR, GB, GR, IE, IT, LU, MC, NL, PT, SE				
	WO 9829388	A1	19980709	WO 1997-CA1013	19971230
	W: CA, JP, US				
	EP 889863	A2	19990113	EP 1997-951051	19971230
	EP 889863	B1	20030507		
	R: DE, FR, GB, IT				
	EP 890176	A1	19990113	EP 1997-951052	19971230
	EP 890176	B1	20010620		
	R: DE, FR, GB, IT				
	JP 2000508114	T2	20000627	JP 1998-529517	19971230
	JP 2000508346	T2	20000704	JP 1998-529516	19971230
	JP 2000508676	T2	20000711	JP 1998-529514	19971230
	JP 2000508677	T2	20000711	JP 1998-529515	19971230
	JP 2000508678	T2	20000711	JP 1998-529518	19971230
	JP 2002514245	T2	20020514	JP 1998-529513	19971230
	US 6120696	A	20000919	US 1998-125792	19980828
	US 6171522	B1	20010109	US 1998-101811	19981119
	US 6333425	B1	20011225	US 1998-101810	19981119
	US 6228942	B1	20010508	US 1998-125798	19981202
	US 6395367	B1	20020528	US 1998-125799	19981202
	US 6319428	B1	20011120	US 1998-125797	19981203
	US 6365068	B1	20020402	US 2000-609362	20000630
	US 6576159	B1	20030610	US 2000-638793	20000809
	US 2001024749	A1	20010927	US 2001-826941	20010406
	US 6506517	B2	20030114		
	US 2002009650	A1	20020124	US 2001-858439	20010516
	US 2002102380	A1	20020801	US 2002-107742	20020327
	US 6835495	B2	20041228		
	US 2003052310	A1	20030320	US 2002-253035	20020924
	US 2003066988	A1	20030410	US 2002-253970	20020924
	US 2005074668	A1	20050407	US 2004-789453	20040227
	US 2005123831	A1	20050609	US 2004-926283	20040825
PRAI	CA 1996-2194127	A	19961230		

CA 1997-2199231	A	19970305
WO 1997-CA1008	W	19971230
WO 1997-CA1009	W	19971230
WO 1997-CA1010	W	19971230
WO 1997-CA1011	W	19971230
WO 1997-CA1012	W	19971230
WO 1997-CA1013	W	19971230
US 1998-101810	A3	19981119
US 1998-101811	A3	19981119
US 1998-125798	A3	19981202
US 1998-125799	A3	19981202
US 1998-125797	A1	19981203
US 2000-638793	A1	20000809
US 2001-858439	A1	20010516
US 2002-107742	A1	20020327

OS MARPAT 129:122975

AB The salts comprise a cation and R<sub>1</sub>SO<sub>x</sub>N-Z in amts. to balance the pos. and neg. charges, where R<sub>1</sub> is halo, perhaloalkyl (optionally interrupted by O, S, or NH) or -alkaryl, R<sub>2</sub>CF<sub>2</sub>, R<sub>2</sub>CF<sub>2</sub>CF<sub>2</sub>, R<sub>2</sub>CF<sub>2</sub>CF(CF<sub>3</sub>), or CF<sub>3</sub>CFR<sub>2</sub>; R<sub>2</sub> is an organic radical which is not perhalogenated; Z is an electron-withdrawing group, which may be the residue of a polymer or may be a polyvalent group attached to other N-SO<sub>x</sub>R<sub>1</sub> moieties; and x = 1 or 2. Thus, a mixture of 40 mmol acrylonitrile and 60 mmol 4-CH<sub>2</sub>:CHC<sub>6</sub>H<sub>4</sub>SO<sub>2</sub>N-SO<sub>2</sub>CF<sub>3</sub> Li<sup>+</sup> was copolymd. in 82% yield by use of 1,1'-azobis(cyclohexanecarbonitrile) in THF, and the copolymer was used at 20% concentration as a binder in both the carbon anode

and the carbon-LiNiO<sub>2</sub> cathode of a battery containing a gelled electrolyte.

IC ICM C07C311-48

ICS C07C311-09; C07D307-64; C07D303-34; C07D407-04; C07D207-452; C07D213-76; C07D285-135; C07D251-70; C07D219-10; C07D311-82

CC 35-3 (Chemistry of Synthetic High Polymers)

Section cross-reference(s): 52, 67, 76

ST sulfonamide salt ionic conductor

IT Cathodes

(containing salts of perfluorinated sulfonamides or sulfinamides)

IT Extrusion of plastics and rubbers

(of poly(ethylene oxide) films containing salts of perfluorinated sulfonamides or sulfinamides)

IT Secondary batteries

(polymeric salts of perfluorinated sulfonamides or sulfinamides for use in)

IT Aldol condensation catalysts

Friedel-Crafts reaction catalysts

Michael reaction catalysts

Polymerization catalysts

(salts of perfluorinated sulfonamides or sulfinamides for use as)

IT Conducting polymers

(salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT Polyoxyalkylenes, reactions

RL: RCT (Reactant); RACT (Reactant or reagent)

(salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT Polyoxyalkylenes, preparation

Polysiloxanes, preparation

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)

(salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT Diels-Alder reaction catalysts

(stereospecific; salts of perfluorinated sulfonamides or sulfinamides for use as)

IT 78-94-4, 3-Buten-2-one, reactions 542-92-7, Cyclopentadiene, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides as Diels-Alder reaction catalysts for)

IT 5063-03-6P  
 RL: PNU (Preparation, unclassified); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides as Diels-Alder reaction catalysts for preparation of)

IT 100-66-3, Anisole, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides as Friedel-Crafts acylation catalysts for)

IT 100-06-1P, p-Methoxyacetophenone  
 RL: PNU (Preparation, unclassified); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides as Friedel-Crafts acylation catalysts for preparation of)

IT 94-41-7, Chalcone  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides as Michael addition catalysts for)

IT 58649-05-1P  
 RL: PNU (Preparation, unclassified); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides as Michael addition catalysts for preparation of)

IT 100-52-7, Benzaldehyde, reactions 31469-15-5  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides as aldol condensation catalysts for)

IT 35022-33-4P  
 RL: PNU (Preparation, unclassified); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides as aldol condensation catalysts for preparation of)

IT 89183-45-9, Polyaniline hydrochloride  
 RL: POF (Polymer in formulation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides as dopants for)

IT 210291-18-2 210291-20-6 210291-21-7  
 RL: CAT (Catalyst use); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as Diels-Alder reaction catalysts)

IT 210291-16-0 210291-17-1  
 RL: CAT (Catalyst use); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as catalysts)

IT 79509-46-9P, Poly(1,3,4-thiadiazole-2,5-diyldithio)  
 RL: BYP (Byproduct); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-23-9P 210291-14-8P  
 RL: CAT (Catalyst use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-17-1P  
 RL: DEV (Device component use); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-35-3P 210227-74-0P

RL: DEV (Device component use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 25038-76-0P, Norbornene homopolymer  
 RL: IMF (Industrial manufacture); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-28-4P  
 RL: MOA (Modifier or additive use); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-43-3  
 RL: PRP (Properties)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-20-6P  
 RL: PRP (Properties); RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-09-1P 210227-16-0P 210227-31-9P 210227-37-5P  
 210227-72-8P  
 RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 210227-39-7P 210227-68-2P  
 RL: PRP (Properties); SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

IT 51-79-6, Ethyl carbamate 62-53-3, Benzenamine, reactions 74-89-5, Methylamine, reactions 78-08-0, Vinyltrietoxysilane 92-82-0, Phenazine 95-54-5, o-Phenylenediamine, reactions 96-24-2, 3-Chloro-1,2-propanediol 97-93-8, Triethylaluminum, reactions 98-16-8, 3-(Trifluoromethyl)aniline 98-61-3, 4-Iodobenzenesulfonyl chloride 102-54-5, Ferrocene 111-92-2, Dibutylamine 142-84-7, Dipropylamine 143-15-7, Dodecyl bromide 354-64-3, Pentafluoroethyl iodide 358-23-6, Trifluoromethanesulfonic anhydride 375-72-4, Perfluorobutane-1-sulfonyl fluoride 392-95-0, 2-Chloro-3,5-dinitrobenzotrifluoride 421-83-0, Trifluoromethanesulfonyl chloride 541-59-3, Maleimide 581-28-2, 2-Aminoacridine 605-65-2, 5-(Dimethylamino)-1-naphthalenesulfonyl chloride 700-16-3, Pentafluoropyridine 764-48-7, Ethylene glycol monovinyl ether 814-68-6, Acryloyl chloride 917-54-4, Methylolithium 920-66-1, 1,1,1,3,3,3-Hexafluoro-2-propanol 1070-89-9, Sodium bis(trimethylsilyl)amide 1111-78-0, Ammonium carbamate 1120-71-4, 1,3-Propane sultone 1120-99-6, 1,2,4-Triazin-3-amine 1126-79-0, Butoxybenzene 1622-32-8, 2-Chloroethanesulfonyl chloride 1633-82-5, 3-Chloropropane-1-sulfonyl chloride 1648-99-3, 2,2,2-Trifluoroethanesulfonyl chloride 2444-68-0, 9-Vinylanthracene 2495-39-8 2633-67-2, 4-Styrenesulfonyl chloride 3520-42-1, Sulforhodamine B 4286-55-9, 6-Bromo-1-hexanol 4628-94-8, Dipotassium 1,3,4-thiadiazole-2,5-dithiolate 5130-24-5, Vinyl chloroformate 5231-87-8 6553-96-4, 2,4,6-Triisopropylbenzenesulfonyl chloride 7673-09-8, Trichloromelamine 7795-95-1, 1-Octanesulfonyl chloride 9036-19-5, Igepal CA 520 10444-89-0 10531-50-7, (R)-2,2,2-Trifluoro-1-phenylethanol 13036-75-4, Fluorosulfonic anhydride 13360-57-1, Dimethylsulfamoyl chloride 13781-67-4, 3-Thiopheneethanol 20611-81-8,

Disodium cyanamide 21797-13-7 : 25322-68-3 27835-99-0 40724-67-2  
 82985-35-1, Bis[3-(trimethoxysilyl)propyl]amine 210226-82-7  
 210227-12-6, 3-(1,1,2,2-Tetrafluoroethoxy)benzenesulfonyl chloride  
 210227-69-3  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic  
 conductors and as catalysts)

IT 421-85-2P, Trifluoromethanesulfonamide 10438-96-7P 14418-84-9P,  
 2-Propenesulfonyl chloride 14986-54-0P, Sulfamoyl fluoride 30334-69-1P  
 31795-44-5P, Sodium 5-formyl-2-furansulfonate 41804-89-1P, Potassium  
 trifluoromethanesulfinate 64773-40-6P, Pentafluoroethanesulfonyl  
 chloride 78491-70-0P 124863-24-7P 162134-09-0P 210226-81-6P,  
 1,2,4-Triazine-3-sulfonyl chloride 210226-83-8P 210226-84-9P  
 210226-85-0P 210226-86-1P 210226-87-2P 210226-88-3P 210226-89-4P  
 210226-91-8P 210226-93-0P 210227-24-0P 210227-38-6P 210227-58-0P  
 210227-71-7P 210227-73-9P 210227-76-2P  
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT  
 (Reactant or reagent)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic  
 conductors and as catalysts)

IT 23384-11-4P, N-[3-(Trifluoromethyl)phenyl]trifluoromethanesulfonamide  
 35534-15-7P 51903-48-1P 152894-19-4P 210226-80-5P 210226-90-7P  
 210226-92-9P 210226-94-1P 210226-95-2P 210226-97-4P 210226-98-5P  
 210226-99-6P 210227-00-2P 210227-01-3P 210227-02-4P 210227-04-6P  
 210227-08-0P 210227-11-5P 210227-13-7P 210227-14-8P 210227-15-9P  
 210227-19-3P 210227-21-7P 210227-26-2P 210227-27-3P 210227-32-0P  
 210227-33-1P 210227-36-4P 210227-41-1P 210227-42-2P 210227-44-4P  
 210227-45-5P 210227-49-9P 210227-51-3P 210227-52-4P 210227-59-1P  
 210227-60-4P 210227-64-8P 210227-65-9P 210227-66-0P 210227-67-1P  
 210227-70-6P 210291-13-7P 210291-15-9P 210304-78-2P  
 RL: SPN (Synthetic preparation); PREP (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic  
 conductors and as catalysts)

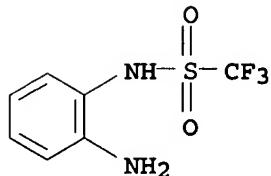
IT 421-85-2DP, Trifluoromethanesulfonamide, reaction products with  
 (chloromethylene)dimethylammonium chloride and poly(sodium  
 p-styrenesulfonate) 3724-43-4DP, (Chloromethylene)dimethylammonium  
 chloride, reaction products with poly(sodium p-styrenesulfonate) and  
 trifluoromethanesulfonamide 25704-18-1DP, Poly(sodium  
 p-styrenesulfonate), reaction products with (chloromethylene)dimethylammonium  
 chloride and trifluoromethanesulfonamide 156118-35-3DP,  
 Dimethylsilanediol-methylsilanediol copolymer, reaction products with  
 lithiated N-(trifluoromethanesulfonyl)vinylsulfonamide 210227-05-7P  
 210227-06-8P 210227-07-9P 210227-10-4P 210227-29-5P 210227-46-6P  
 210227-47-7P 210227-50-2P 210227-55-7P 210227-57-9P 210227-62-6P  
 210227-63-7P 210227-79-5P 210227-81-9P 210227-82-0P  
 210227-84-2P 210227-85-3DP, reaction products with di-Me, Me hydrogen  
 polysiloxane 210291-10-4P 210291-12-6P  
 RL: SPN (Synthetic preparation); TEM (Technical or engineered material  
 use); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic  
 conductors and as catalysts)

IT 210227-72-8P  
 RL: PRP (Properties); SPN (Synthetic preparation); PREP  
 (Preparation)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic  
 conductors and as catalysts)

RN 210227-72-8 HCPLUS

CN Methanesulfonamide, N-(2-aminophenyl)-1,1,1-trifluoro-, homopolymer (9CI)  
 (CA INDEX NAME)

CM 1

CRN 53718-45-9  
CMF C7 H7 F3 N2 O2 S

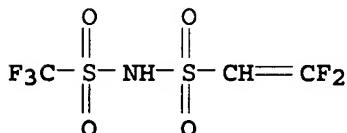
IT 210227-82-0P

RL: SPN (Synthetic preparation); TEM (Technical or engineered material use); PREP (Preparation); USES (Uses)  
 (salts of perfluorinated sulfonamides or sulfinamides for use as ionic conductors and as catalysts)

RN 210227-82-0 HCPLUS

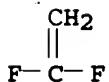
CN Ethenesulfonamide, 2,2-difluoro-N-[(trifluoromethyl)sulfonyl]-, lithium salt, polymer with 1,1-difluoroethene (9CI) (CA INDEX NAME)

CM 1

CRN 210227-15-9  
CMF C3 H2 F5 N O4 S2 . Li

● Li

CM 2

CRN 75-38-7  
CMF C2 H2 F2

L32 ANSWER 16 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN  
 AN 1997:627049 HCPLUS  
 DN 127:294058  
 TI Synthesis of a hexafluoropropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole  
 AU Selampinar, Fatma; Akbulut, Ural; Yilmaz, Tulay; Gungor, Attila; Toppore,

Levent  
CS Department of Chemistry, Middle East Technical University, Ankara, 06531, Turk.  
SO Journal of Polymer Science, Part A: Polymer Chemistry (1997), 35(14), 3009-3016  
CODEN: JPACEC; ISSN: 0887-624X  
PB Wiley  
DT Journal  
LA English  
AB A new elec. conducting composite film from polypyrrole and 4,4'-(hexafluoroisopropylidene)-bis(phthalic anhydride)-based polyimide was prepared. Pyrrole and the dopant ion can easily penetrate through the polyimide substrate and electropolymerize on the platinum (Pt) electrode due to the swelling of the polyimide on the metal electrode. The electrochem. properties of polypyrrole-polyimide (PPy/PI) composite films have been investigated by using cyclic voltammetry. The PPy/PI composite film is suitable for use as the electroactive material owing to its stable and controllable electrochem. properties. The elec. conductivity of composites falls in the range 0.0035-15 S/cm. Scanning electron micrograph, FTIR, and thermal studies indicate that PPy and PI form a homogeneous material rather than a simple mixture  
CC 37-6 (Plastics Manufacture and Processing)  
Section cross-reference(s): 76  
ST fluoropolymer polyimide blend polypyrrole elec cond; electrochem polymn pyrrole fluoropolymer polyimide matrix  
IT Polymerization  
(electrochem.; in preparation of polypyrrole and its conducting polymer composites with hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide)  
IT Conducting polymers  
Electric conductivity  
Oxidation, electrochemical  
Polymer morphology  
Reduction, electrochemical  
(hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)  
IT Polymer blends  
RL: PRP (Properties); SPN (Synthetic preparation); PREP (Preparation)  
(hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)  
IT Dopants  
(hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole containing)  
IT Polysulfones, preparation  
Polysulfones, preparation  
Polysulfones, preparation  
Polysulfones, preparation  
RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
(polyether-polyimide-, fluorine-containing; hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole containing)  
IT Fluoropolymers, preparation  
RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
(polyether-polyimide-polysulfone-; hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole containing)  
IT Polyimides, preparation  
Polyimides, preparation

Polyimides, preparation  
 Polyimides, preparation  
 RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (polyether-polysulfone-, fluorine-containing; hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole containing)

IT Polyethers, preparation  
 Polyethers, preparation  
 Polyethers, preparation  
 Polyethers, preparation  
 RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (polyimide-polysulfone-, fluorine-containing; hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole containing)

IT 429-42-5, Tetrabutylammonium tetrafluoroborate  
 RL: MOA (Modifier or additive use); USES (Uses)  
 (dopant; hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)

IT 30604-81-0P, Polypyrrole 118085-79-3P 118087-85-7P  
 RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)

IT 30203-11-3P  
 RL: RCT (Reactant); SPN (Synthetic preparation); PREP (Preparation); RACT (Reactant or reagent)  
 (monomer; synthesis of a hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)

IT 80-07-9, Bis(4-chlorophenyl) sulfone 591-27-5, 3-Aminophenol  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (reactant in monomer synthesis; synthesis of a hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)

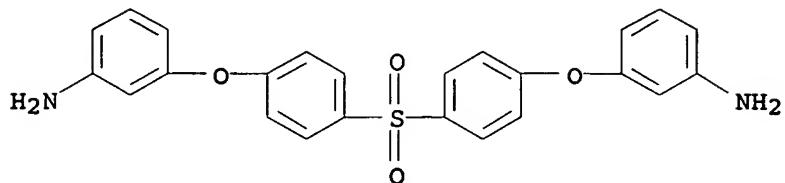
IT 118085-79-3P  
 RL: POF (Polymer in formulation); PRP (Properties); SPN (Synthetic preparation); PREP (Preparation); USES (Uses)  
 (hexafluoroisopropylidene-bis(phthalic anhydride)-based polyimide and its conducting polymer composites with polypyrrole)

RN 118085-79-3 HCPLUS  
 CN 1,3-Isobenzofurandione, 5,5'-[2,2,2-trifluoro-1-(trifluoromethyl)ethylidene]bis-, polymer with 3,3'-[sulfonylbis(4,1-phenyleneoxy)]bis[benzenamine] (9CI) (CA INDEX NAME)

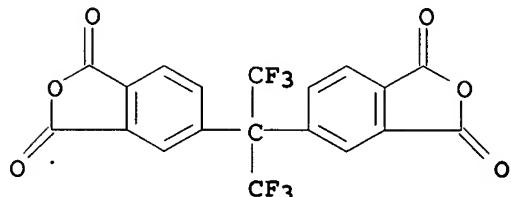
CM 1

CRN 30203-11-3

CMF C24 H20 N2 O4 S



CM 2

CRN 1107-00-2  
CMF C19 H6 F6 O6RE.CNT 21 THERE ARE 21 CITED REFERENCES AVAILABLE FOR THIS RECORD  
ALL CITATIONS AVAILABLE IN THE RE FORMATL32 ANSWER 17 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN  
AN 1996:666523 HCAPLUS

DN 125:288712

TI Manufacture of antistatic-layered support for silver halide photographic material having improved stability

IN Tachibana, Noriki; Kotani, Chiaki; Okamura, Shinichi; Morita, Seiwa

PA Konishiroku Photo Ind., Japan

SO Jpn. Kokai Tokkyo Koho, 56 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI JP 08201978	A2	19960809	JP 1995-8995	19950124
PRAI JP 1995-8995		19950124		

AB The method is characterized by that the amount of exposure (product of light intensity and exposing time) received by the support between the time interval of the processes of (a) forming on the support an antistatic layer comprising a  $\pi$  electron-conjugated elec. conductive polymer (absorption maximum in 300-700 nm) and (b) coating of the Ag halide emulsion layers does not exceed 15,000 lx-h. Preferable  $\pi$  electron-conjugated polymers are polypyrrole, polythiophene, polyfuran, polyaniline, poly(isothianaphthene), poly-p-phenylene-vinylidene, and their derivs. and copolymers. By limiting the exposure below the above amount, the antistatic property is maintained throughout the storage period of the support, which improves the consistency of photog. property.

IC ICM G03C001-85

ICS G03C005-08

CC 74-2 (Radiation Chemistry, Photochemistry, and Photographic and Other Reprographic Processes)

Section cross-reference(s): 38, 76

ST elec conductive polymer antistatic photog support; polyester film antistatic photog support

IT Electric conductors, polymeric  
Photographic films  
(manufacture of antistatic-layered support for silver halide photog. material having improved stability)

IT Poly(arylenealkenylenes)  
Polyamines  
Polyoxyphenylenes  
Polythiophenylenes  
RL: DEV (Device component use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)  
(manufacture of antistatic-layered support for silver halide photog. material having improved stability)

IT 429-06-1, Tetraethylammonium tetrafluoroborate 429-42-5, Tetrabutylammonium tetrafluoroborate 1493-13-6, Trifluoromethanesulfonic acid 2926-30-9, Sodium trifluoromethanesulfonate 7550-35-8, Lithium bromide 7553-56-2, Iodine, uses 7647-01-0, Hydrogen chloride, uses 7664-93-9, Sulfuric acid, uses 7705-08-0, Iron trichloride, uses 7783-70-2, Antimony pentafluoride 7784-36-3, Arsenic pentafluoride 7789-21-1, Fluorosulfuric acid 7791-03-9, Lithium perchlorate 14075-53-7, Potassium tetrafluoroborate 14283-07-9, Lithium tetrafluoroborate  
RL: MOA (Modifier or additive use); USES (Uses)  
(conductive polymer dopant; manufacture of antistatic-layered support for silver halide photog. material having improved stability)

IT 83560-37-6, Poly(thio-1,2-ethenediyl)  
RL: DEV (Device component use); USES (Uses)  
(manufacture of antistatic-layered support for silver halide photog. material having improved stability)

IT 98-70-4DP, polymers with poly(arylenealkenylenes) 25190-54-9P  
25233-30-1P 25233-34-5P, Polythiophene 26009-24-5DP,  
Poly(p-phenylenevinylene), polymers with sulfonated styrene derivs.  
26498-02-2P, Poly(2,5-thiophenediyl-1,2-ethenediyl) 27073-41-2P  
27082-18-4P 30604-81-0P, Polypyrrole 33411-63-1P 89761-73-9P  
91201-85-3P 94750-56-8P 95831-23-5P 97126-62-0P 99742-70-8P  
105935-08-8P 110847-38-6P 114267-74-2P 118337-98-7P  
121536-25-2P 122721-92-0P 132670-08-7P 162369-94-0P 162369-98-4P  
162370-00-5P 165455-34-5P 181226-79-9P 181226-81-3P 181226-82-4P  
181226-85-7P 181226-86-8P 181226-87-9P 181226-88-0P 182956-05-4P  
182956-11-2P 182956-12-3P 182956-13-4P 182956-14-5P  
182956-15-6P 182956-19-0P 182956-23-6P 182956-25-8P  
182956-28-1P  
RL: DEV (Device component use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)  
(manufacture of antistatic-layered support for silver halide photog. material having improved stability)

IT 118337-98-7P 182956-15-6P  
RL: DEV (Device component use); IMF (Industrial manufacture); PREP (Preparation); USES (Uses)  
(manufacture of antistatic-layered support for silver halide photog. material having improved stability)

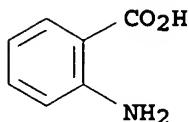
RN 118337-98-7 HCPLUS

CN Benzoic acid, 2-amino-, homopolymer (9CI) (CA INDEX NAME)

CM 1

CRN 118-92-3

CMF C7 H7 N O2



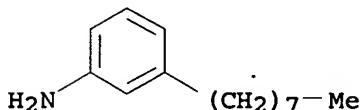
RN 182956-15-6 HCPLUS

CN Benzenesulfonic acid, 4-ethenyl-, polymer with 3-octylbenzenamine (9CI)  
(CA INDEX NAME)

CM 1

CRN 118198-99-5

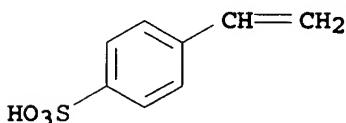
CMF C14 H23 N



CM 2

CRN 98-70-4

CMF C8 H8 O3 S



L32 ANSWER 18 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 1995:17082 HCPLUS

DN 122:32239

TI Electrosynthesis of conducting polyaniline in aqueous HF

AU Bhadani, S. N.; Sen Gupta, S. K.; Gupta, M. K.; Kumari, M.

CS Dep. Chem., Ranchi Univ., Ranchi, 834 008, India

SO Journal of Polymer Materials (1993), 10(2), 117-22

CODEN: JOPME8; ISSN: 0970-0838

DT Journal

LA English

AB Polyaniline films were synthesized on Pt sheets in aqueous solution of aniline and HF under potentiostatic conditions at +0.8 V vs SCE. The polymeric film synthesized at low temperature and high acid concentration exhibits high elec. conductivities. The cyclic voltammograms of polyaniline were examined in the range -0.2 V to +1.0 V vs SCE in the presence and absence of aniline in the aqueous solution of HF. Cyclic voltammetry investigations show that the polymeric film suffers degradation when

potential exceeds +0.8 V and below this potential it is quite stable. The mechanism for electrochem. redox reaction is suggested.

CC 35-7 (Chemistry of Synthetic High Polymers)

ST polyaniline electropolymer hydrogen fluoride; mechanism electropolymer aniline

IT Electric conductivity and conduction (elec. conductivity of polyaniline prepared by electropolymer. in aqueous HF)

IT Polymerization (mechanism of electropolymer. of aniline in aqueous HF)

IT Polyamines

RL: SPN (Synthetic preparation); PREP (Preparation) (aniline-based, electrosynthesis of conducting polyaniline in aqueous HF)

IT 7664-39-3, Hydrogen fluoride, uses

RL: NUU (Other use, unclassified); USES (Uses) (electrosynthesis of conducting polyaniline in aqueous HF)

IT 25233-30-1P, Polyaniline

RL: SPN (Synthetic preparation); PREP (Preparation) (electrosynthesis of conducting polyaniline in aqueous HF)

IT 62-53-3, Aniline, processes

RL: PEP (Physical, engineering or chemical process); PROC (Process) (mechanism of electropolymer. of aniline in aqueous HF)

L32 ANSWER 19 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1993:201166 HCAPLUS

DN 118:201166

TI Discharge and charge characteristics of polyaniline prepared by electropolymerization of aniline in nonaqueous solvent

AU Yonezawa, Susumu; Kanamura, Kiyoshi; Takehara, Zenichiro

CS Fac. Eng., Kyoto Univ., Kyoto, 606, Japan

SO Journal of the Electrochemical Society (1993), 140(3), 629-33  
CODEN: JESOAN; ISSN: 0013-4651

DT Journal

LA English

AB Electroactive polyaniline can be prepared in nonaq. solns. (such as propylene carbonate (PC), MeCN) containing 1.0 mol dm<sup>-3</sup> aniline tetrafluoroborate (ATFB). The polyaniline prepared in PC containing 1.0 mol dm<sup>-3</sup> ATFB showed a high discharge capacity compared with that prepared in 2.0 mol dm<sup>-3</sup> HBF<sub>4</sub> aqueous solution containing 1.0 mol dm<sup>-3</sup> aniline (pH 1) because oxidative degradation during the electropolymer. did not occur in PC containing 1.0 mol dm<sup>-3</sup> ATFB. However, the decrease in the discharge capacity of polyaniline prepared in PC containing 1.0 mol dm<sup>-3</sup> ATFB with increasing discharge current was larger than that prepared in 2.0 mol dm<sup>-3</sup> HBF<sub>4</sub> aqueous solution containing 1.0 mol dm<sup>-3</sup> aniline (pH 1). From SEM observations, it can be seen that these polyanilines have different morphologies.

CC 72-2 (Electrochemistry)

Section cross-reference(s): 35, 36, 52

ST aniline electropolymer polyaniline charge discharge characteristic; battery electrode polyaniline charge discharge characteristic; solvent effect electropolymer aniline; redox reaction electrochem polyaniline film; fluoroborate aniline electropolymer nonaq soln

IT Polymer morphology (of polyaniline film on platinum)

IT Solvent effect (on electrochem. polymerization of aniline and on charge and

: discharge characteristics)  
 IT Electric conductors, polymeric  
 (polyaniline, electrochem., preparation and charge and discharge  
 characteristics of)  
 IT Cathodes  
 (battery, polyaniline)  
 IT Electrodes  
 (battery, polyaniline, characteristics of)  
 IT Polymerization  
 (electrochem., of aniline on platinum in nonaq. solns. and of  
 aniline tetrafluoroborate on platinum in acetonitrile and in  
 propylene carbonate)  
 IT Redox reaction  
 (electrochem., of polyaniline film on platinum)  
 IT 108-32-7, Propylene carbonate  
 RL: PRP (Properties)  
 (discharge capacity of polyaniline electrode prepared in  
 various electrolytes in lithium tetrafluoroborate-containing)  
 IT 14283-07-9, Lithium tetrafluoroborate(1-)  
 RL: PRP (Properties)  
 (discharge capacity of polyaniline electrode prepared in  
 various electrolytes in propylene carbonate containing)  
 IT 16872-11-0, Hydrogen tetrafluoroborate(1-)  
 RL: PRP (Properties)  
 (electrochem. polymerization of aniline on platinum in aqueous  
 solution containing aniline and)  
 IT 3796-29-0, Aniline tetrafluoroborate(1-)  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (electrochem. polymerization of, on platinum in acetonitrile and in  
 propylene  
 carbonate, charge-discharge characteristics of polyaniline  
 electrode in relation to)  
 IT 25233-30-1, Polyaniline  
 RL: PRP (Properties)  
 (electrochem. preparation and discharge and charge characteristics of  
 electrodes of)  
 IT 7440-06-4, Platinum, uses  
 RL: USES (Uses)  
 (electrode, polymerization of aniline on, in aqueous and  
 nonaq. solns.)  
 IT 62-53-3, Aniline, reactions  
 RL: RCT (Reactant); RACT (Reactant or reagent)  
 (polymerization of, electrochem., in aqueous aniline and  
 tetrafluoroboric acid solns.)

L32 ANSWER 20 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 1989:155779 HCPLUS

DN 110:155779

TI Manufacture of electrically conductive polymer  
 compositions

IN Sakai, Toshiyuki; Kobayashi, Masao

PA Showa Denko K. K., Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
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PI JP 63215772	A2	19880908	JP 1987-47776	19870304

JP 06078493. B4 19941005

PRAI JP 1987-47776 19870304

AB Title compns. are manufactured by polymerizing monomers capable of forming anionic

polymer electrolytes in the presence of polymers of conjugated  $\pi$ -electron structure. The compns. are dispersions of the anionic polymer electrolytes in the polyconjugated polymers and maintain elec. conductivity over a prolonged period of time. Thus, isothianaphthene was electrochem. polymerized in MeCN containing Ph<sub>4</sub>PCl to form polyisothianaphthene (I) on an In oxide-Sn oxide-coated glass electrode; I was dedoped and freed from the electrolyte, immersed in an aqueous solution of vinylsulfuric acid (II), ultrasonically treated to obtain a uniform dispersion of II in I, taken out, and irradiated with long-wave UV in air for 5 min. The electrode containing a composition of I and poly(vinylsulfuric acid) was kept at +0.5 V for 1 min in MeCN containing Et<sub>4</sub>NClO<sub>4</sub> with a Pt plate as opposing electrode and Ag/Ag<sup>+</sup> as reference electrode to give a transparent, homogeneous polymer composition, which remained transparent when freed from the electrolyte, dried in vacuo, and stored in air for 50 days.

IC ICM C08L101-00

ICS C08F002-44

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 37, 76

ST polyconjugated polymer anionic polymer blend; elec conductive polymer compn; polyisothianaphthene elec cond

IT Electric conductors

(blends of polyconjugated polymers and anionic polymer electrolytes, manufacture of)

IT Polyphosphoric acids

RL: USES (Uses)

(blends with conjugated polymers, elec. conductive, manufacture of)

IT Fluoropolymers

RL: PREP (Preparation)

(carboxy-containing, blends with polyconjugated polymers, elec. conductive, manufacture of)

IT Sulfonic acids, polymers

RL: USES (Uses)

(fluoro, polymers, blends with polyconjugated polymers, elec. conductive, manufacture of)

IT Fluoropolymers

RL: PREP (Preparation)

(sulfo-containing, blends with polyconjugated polymers, elec. conductive, manufacture of)

IT 25067-58-7P, Polyacetylene 25190-62-9P, Poly(p-phenylene) 25233-30-1P,

Polyaniline 25233-34-5P, Polythiophene 30604-81-0P,

Polypyrrole 91201-85-3P, Poly(isothianaphthene)

RL: PREP (Preparation)

(blends with anionic polymer electrolytes, elec. conductive, manufacture of)

IT 9003-01-4P, Acrylic acid homopolymer 25087-26-7P, Methacrylic acid

homopolymer 25191-25-7P 25513-46-6P, Glutamic acid homopolymer

25608-40-6P, Aspartic acid homopolymer 26101-52-0P, Vinylsulfonic acid homopolymer 50851-57-5P 119959-66-9P

RL: PREP (Preparation)

(blends with polyconjugated polymers, elec. conductive, manufacture of)

L32 ANSWER 21 OF 23 HCPLUS COPYRIGHT 2006 ACS on STN

AN 1989:11095 HCPLUS

DN 110:11095  
 TI Secondary nonaqueous battery  
 IN Toyosawa, Shinichi; Kijima, Shigeru; Daifuku, Hideharu; Maeda, Yuko; Arai, Katsuhiko; Kawagoe, Takahiro  
 PA Bridgestone Corp., Japan  
 SO Ger. Offen., 15 pp.  
 CODEN: GWXXBX  
 DT Patent  
 LA German  
 FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	DE 3808985	A1	19880929	DE 1988-3808985	19880317
	JP 63230706	A2	19880927	JP 1987-64995	19870318
	JP 63258936	A2	19881026	JP 1987-93962	19870416
	JP 63289765	A2	19881128	JP 1987-124921	19870521
	JP 63289766	A2	19881128	JP 1987-124922	19870521
	FR 2612695	A1	19880923	FR 1988-3459	19880317
	US 4906538	A	19900306	US 1988-169881	19880318
	US 4904553	A	19900227	US 1988-211059	19880624
PRAI	JP 1987-64995	A	19870318		
	JP 1987-93962	A	19870416		
	JP 1987-124921	A	19870521		
	JP 1987-124922	A	19870521		
	JP 1987-164968	A	19870701		
	JP 1987-238168	A	19870922		
	JP 1987-249146	A	19871002		
	JP 1987-292391	A	19871119		
	JP 1987-306715	A	19871203		
	JP 1988-29707	A	19880210		
	US 1988-169881	A2	19880318		

AB The battery has a cathode of polyaniline containing 15-30 weight% BF<sub>4</sub><sup>-</sup>, an Al-(25-65 atomic%) Li alloy anode, and an electrolyte of >1 to 3M LiBF<sub>4</sub> in a MeOC<sub>2</sub>H<sub>4</sub>OMe-(35-65 volume%) propylene carbonate mixture. The polyaniline is prepared by electrolytic polymerization using a stainless steel electrode and a segmented counterelectrode. The polyaniline film is purified by flowing H<sub>2</sub>O or an organic solvent through the film in its thickness direction. The extract obtained by extracting 1 g polyaniline with 200 mL H<sub>2</sub>O has an equilibrium pH value of 2.4-4. A button-type battery of the invention had a capacity of 4.9 mA-h and retained 86% of its initial capacity after 1-wk storage at 60° vs. 2.7 mA-h and 76% and 3.0 mA-h and 62% for batteries having polyaniline cathodes containing 14 and 36 weight% BF<sub>4</sub><sup>-</sup>, resp.

IC ICM H01M004-60  
 ICS H01M006-14

ICA C08L079-00; C08G073-00; C25B003-10

CC 52-2 (Electrochemical, Radiational, and Thermal Energy Technology)  
 Section cross-reference(s): 38

ST lithium polyaniline nonaq battery; fluoroborate  
 polyaniline battery cathode

IT Electric conductors  
 (polymeric, fluoroborate-containing polyaniline  
 , for secondary lithium batteries)

IT Batteries, secondary  
 (button-type, aluminum lithium alloy-polyaniline, with nonaq.  
 electrolyte, containing fluoroborate for high-cycle and storage-life)

IT 12615-39-3 117937-69-6, Aluminum 45-75, lithium 25-65 (atomic)

RL: USES (Uses)  
 (anodes, in nonaq.-electrolyte batteries with fluoroborate-containing  
 polyaniline cathodes)

IT 25233-30-1P, Polyaniline :  
 RL: PREP (Preparation)  
 (fluoroborate-containing, cathodes, manufacture of, for high-cycle and storage-life nonaq. batteries)

IT 14874-70-5  
 RL: USES (Uses)  
 (polyaniline containing, cathodes, manuf of, for high-cycle and storage-life nonaq. batteries)

L32 ANSWER 22 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1988:7040 HCAPLUS

DN 108:7040

TI Dopants for electrically conductive polymers

IN Iwata, Kaoru; Hagiwara, Tsuneo

PA Agency of Industrial Sciences and Technology, Japan

SO Jpn. Kokai Tokkyo Koho, 5 pp.

CODEN: JKXXAF

DT Patent

LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 62119237	A2	19870530	JP 1985-258770	19851120
	JP 01038808	B4	19890816		

PRAI JP 1985-258770 19851120

AB The sulfonates RF<sub>2</sub>(CF<sub>2</sub>)<sub>m</sub>SO<sub>3</sub><sup>-</sup> (R = H, F; m = 3-20) are dopants for elec. conductive, organic polymers. Thus, 200 mL propylene carbonate 0.06M in pyrrole and 0.1M in C<sub>8</sub>F<sub>17</sub>SO<sub>3</sub><sup>-</sup> Et<sub>4</sub>N<sup>+</sup> and 2 mL water were placed in a cell with a Pt anode and Cu cathode and electrolyzed for 26 h at c.d. 5 mA and -20° to give a black, 30-μ film with elec. conductivity 62 S/cm, containing 21 mol% dopant.

IC ICM C08K005-42

ICS C08G061-12; C08G073-00

ICA H01B001-12

CC 37-6 (Plastics Manufacture and Processing)  
 Section cross-reference(s): 76

ST fluoroalkanesulfonate dopant conductive polymer; pyrrole polymer elec conductive doping; doping polymer elec conductor; sulfonate perfluoroalkane doping polymer; perfluoroctanesulfonate doping polymer

IT Electric conductors

(organic polymers doped with fluoroalkanesulfonates as)

IT Sulfonic acids, uses and miscellaneous

RL: USES (Uses)  
 (alkane, perfluoro, doping agents for elec. conductive polymers)

IT 1763-23-1, n-Perfluoroctanesulfonic acid 56773-42-3, Tetraethylammonium n-perfluoroctanesulfonate 60453-92-1 111831-41-5 111864-24-5 111873-33-7

RL: USES (Uses)

(dopant, for elec. conductive organic polymers)

IT 25233-30-1, Polyaniline 30604-81-0, Polypyrrole 84928-92-7, Poly(3-methylthiophene)

RL: USES (Uses)

(elec. conductive, doping of, with perfluoroalkanesulfonates)

L32 ANSWER 23 OF 23 HCAPLUS COPYRIGHT 2006 ACS on STN

AN 1987:139428 HCAPLUS

DN 106:139428

TI Treatment of polymer products  
 IN Ozawa, Shuji; Hagiwara, Tsuneo; Iwata, Kaoru  
 PA Teijin Ltd., Japan  
 SO Jpn. Kokai Tokkyo Koho, 7 pp.  
 CODEN: JKXXAF

DT Patent  
 LA Japanese

FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 61123637	A2	19860611	JP 1984-243409	19841120
PRAI	JP 1984-243409		19841120		

AB Polymer products containing oxidizing agents are treated with organic low-mol. weight compds. which form conductive polymers by oxidative polymerization, to give elec. conductive polymer products. Thus, PVC film containing 50 phr FeCl<sub>3</sub> was soaked 5 h in 0.1 M aqueous pyrrole, washed, and dried to give a black flexible film containing 12.0% polypyrrole and showing conductivity 1.5 + 10<sup>-1</sup> S/cm, vs. ≤10-12 S/cm for PVC.

IC ICM C08J007-12

ICS C08G061-00; H01B001-12

CC 38-3 (Plastics Fabrication and Uses)

Section cross-reference(s): 42, 76

ST elec cond polymer film; PVC film polypyrrole  
elec conductive

IT Polymers, uses and miscellaneous

RL: USES (Uses)

(oxidant-containing, elec. conductive polymer formation on, by  
contacting with pyrrole or thiophene or aniline derivs.)

IT Oxidizing agents

(plastics containing, elec. conductive polymer formation on, by  
contacting with pyrrole or thiophene or aniline derivs.)

IT Electric conductors

(polymers, formation of, on plastic products containing oxidizing agents,  
by contacting with monomer)

IT Films

(elec. conductive, formation of, on plastic products containing  
oxidizing agents, by contacting with monomers)

IT Electric conductors

(films, formation of, on plastic products containing oxidizing  
agents, by contacting with monomers)

IT Polymerization

(oxidative, of pyrrole or thiophene or aniline derivs., elec.  
conductive polymers formation by, on plastic products)IT 79-10-7D, Acrylic acid, esters, polymers 9002-85-1, Poly(vinylidene  
chloride) 9002-86-2, PVC 9002-89-5, Poly(vinyl alcohol) 9003-17-2,  
Polybutadiene 9003-53-6, Polystyrene 9003-56-9, ABS 24981-14-4,  
Vinyl fluoride polymer

RL: USES (Uses)

(oxidant-containing, elec. conductive polymer formation on, by  
contacting with pyrrole or thiophene or aniline derivs.)IT 84-65-1, Anthraquinone 106-51-4, p-Benzoylquinone, uses and miscellaneous  
118-75-2, uses and miscellaneous 130-15-4 7705-08-0, Ferric chloride,  
uses and miscellaneous 13537-24-1, Ferric perchlorate

RL: USES (Uses)

(plastics containing elec conductive polymer formation on, by  
contacting with pyrrole or thiophene or aniline derivs.)

IT 7681-11-0, uses and miscellaneous

RL: USES (Uses)

(plastics containing with iodine, elec. conductive polymer

formation on, by contacting with pyrrole or thiophene or aniline derivs.)

IT 7553-56-2, Iodine, uses and miscellaneous

RL: USES (Uses)

(plastics containing, with potassium iodide, elec. conductive formation on, by contacting with pyrrole or thiophene or aniline derivs.)

IT 25233-30-1P, Polyaniline 30604-81-0P, Polypyrrole

72945-66-5P, Poly(N-methylpyrrole) 79134-59-1P 79799-71-6P,

Poly(N-phenylpyrrole) 84928-92-7P, Poly(3-methylthiophene)

87106-17-0P, Poly(3,4-dimethylpyrrole) 104318-58-3P, Poly(m-toluidine)

RL: PREP (Preparation)

(preparation of conductive, on plastic products containing oxidizing agents)

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